



HOW TO WORK
WITH THE
SPECTROSCOPE



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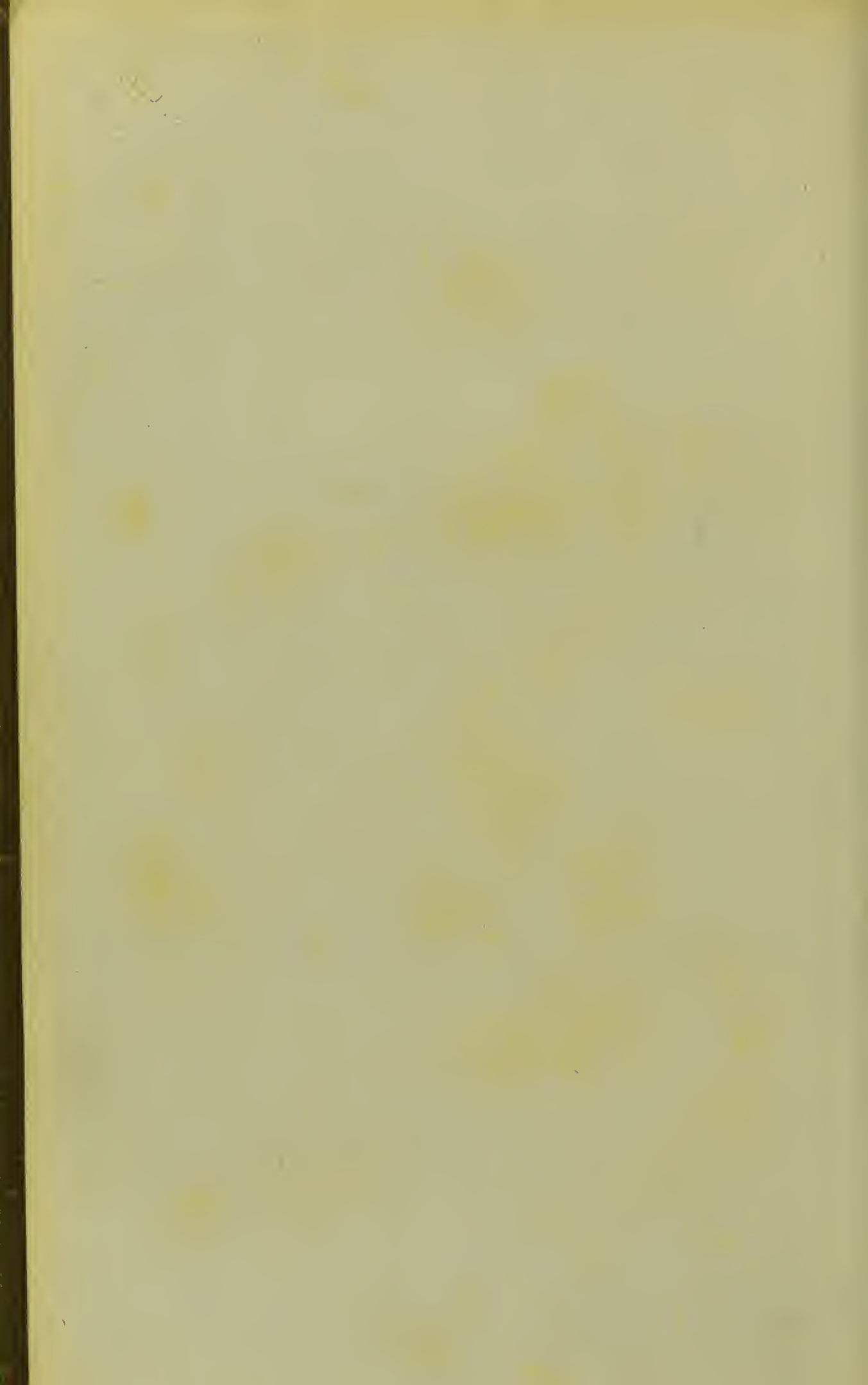
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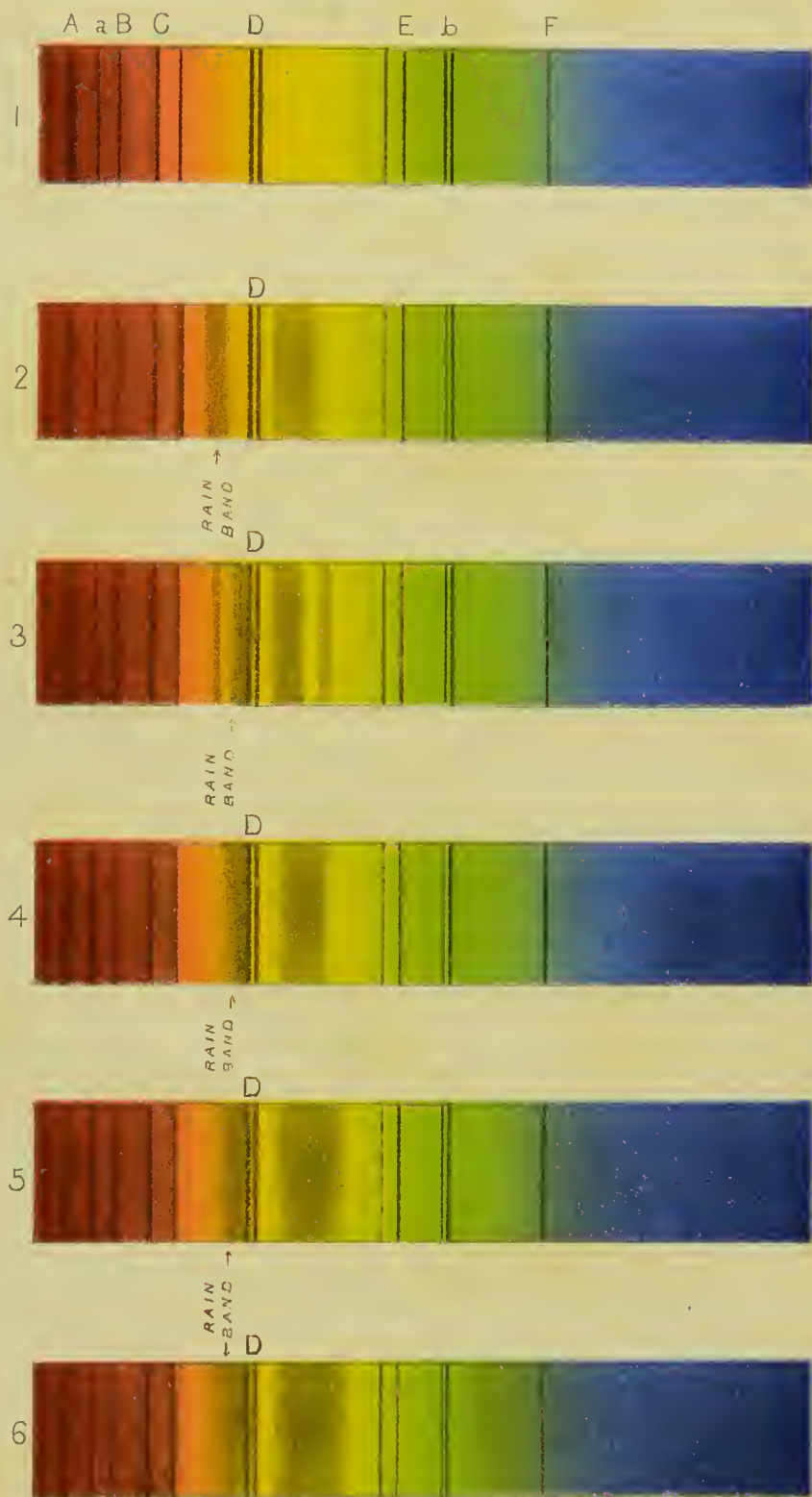




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THE RAINBAND IN SPECTROSCOPE
OF MODERATE DISPERSION.



HOW TO WORK WITH THE SPECTROSCOPE.

A MANUAL

OF

PRACTICAL MANIPULATION WITH SPECTROSCOPES
OF ALL KINDS,

INCLUDING

DIRECT-VISION	SPECTROSCOPES
RAIN-BAND	SPECTROSCOPES
CHEMICAL	SPECTROSCOPES
SOLAR	SPECTROSCOPES
STAR	SPECTROSCOPES
AUTOMATIC	SPECTROSCOPES
MICRO	SPECTROSCOPES
SCREEN	SPECTROSCOPES
BESSEMER	SPECTROSCOPES

And Accessory Apparatus.

WITH ABOVE THIRTY ENGRAVINGS AND DIAGRAMS.

BY

JOHN BROWNING, F.R.A.S., F.R.M.S., M.R.I., &c.

[Signature]

JOHN BROWNING,

63, STRAND, LONDON, W.C.

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P R E F A C E .

FOR many years I have had almost daily to reply to inquiries respecting the best method of manipulating with various kinds of Spectroscopes.

It had been suggested to me long since, by our highest authority on the subject, that I should write a *small* work giving the required information. For reasons which will be understood by my friends, I had a great objection to issue such a work, and trusted it would be done by some of the leading scientific men who have worked in this direction. But recently the inquiries for information have been more and more numerous, and I am compelled, both to save my own time and to assist my numerous correspondents, to attempt to supply what appears to be an evident want.

Mr. Proctor's admirable little Manual on the Work of the Spectroscope will be found invaluable by those who do not wish to incur the expense of Roscoe's or Schellen's works on Spectrum Analysis. But Mr. Proctor states that it did not enter into his plans to give detailed instructions for the use of the various kinds of Spectroscopic Apparatus.

I have, therefore, endeavoured to supply such information in the following Pamphlet. The fact that a List of Prices is appended of the various Instruments which are described, will not, I hope, be considered to detract from its value. My extensive correspondence leads me to conclude that such information is exceedingly welcome to all those who think of making any experiments for themselves. Once provided with Apparatus, the experimentalist should, obtain a copy of Mr. Proctor's book before alluded to, or of Roscoe's or Schellen's larger works on

the subject. Mr. Lockyer's small work on the Spectroscope contains detailed information on the method of working with the Induction Coil, and observing solar prominences.

JOHN BROWNING.

PREFACE TO THE SECOND EDITION.

THIS little book has been out of print several months. I intended to re-write it and make many additions to the contents, but as I cannot find time to do this, and as I have applications daily for the work, I have only added the particulars as to the Rain-band Spectroscope and Mr. Rand Capron's diagram of Rain-band Spectra, which I have had printed in colours. In this condition I trust it may be useful.

December, 1882.

JOHN BROWNING.

SPECTROSCOPES AND SPECTRUM APPARATUS.

HOW TO WORK WITH A SPECTROSCOPE.

First see that the edges of the slit are free from dust. Direct the slit of the instrument towards the sun, the sky, or some bright light. If there are any horizontal lines visible—*that is, lines running parallel with the spectrum*—they are almost sure to be produced by dust. To remove this dust, open the slit as widely as possible, and wipe the edges of the slit with a small wedge of dry wood. Then close the slit completely; re-open it, and the lines will probably have disappeared; if not, repeat the operation. Note that a camel-hair pencil, a leather, cloth, or paper will be sure to *make the slit worse*.

Place the slit as close to the source of light as you can without injuring it.

THE MINIATURE SPECTROSCOPE.

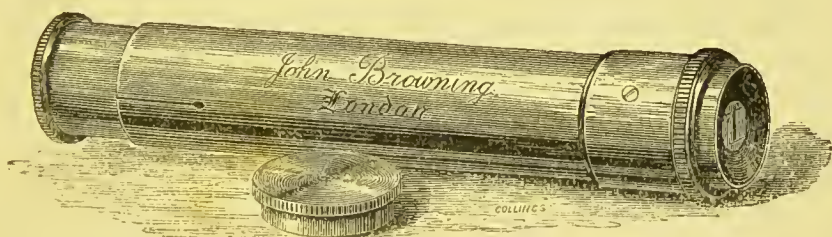


Fig. 1.

This instrument will show many of Fraunhofer's lines, the bright lines of the metals and gases, and the absorption bands in coloured gases, crystals, or liquids.

The Miniature Spectroscope consists of a compound direct-vision prism, which is placed in a sliding drawer. At the end of this drawer there is a lens; in the best instruments this is an achromatic combination. At the opposite end to the eye-piece there is a slit, which, in the instruments of a superior class, is adjustable by a screw motion. The two jaws move equably from the centre, on rotating the milled ring between the thumb and finger. The jaws of this slit will require to be almost closed to view the Fraunhofer lines, but may with advantage be opened wider when viewing the lines of chemical spectra, or absorption bands in coloured liquids.

HOW TO USE THE RAIN-BAND SPECTROSCOPE.



Fig. 2.

For several years past the prediction of the weather has received constantly increasing attention. This makes the introduction of any new method of forecasting the weather of great importance. To nearly every person in the community a means of telling in the morning whether the day will be fine or wet is of considerable value. To farmers and many others whose work goes on exclusively in the open air the power of predicting this is of still greater consequence. The barometer will frequently rise for hours, and yet rain come on and continue falling. As a means of predicting coming rain, no instrument has been introduced which equals the Rain-band Spectroscope.

To use the Rain-band Spectroscope, close the jaws of the slit as closely as possible without making them touch. This is done by turning the second milled ring on the smaller end of the Spectroscope. If shut too closely—that is, the jaws are made to touch—strong black lines will appear *parallel* to the spectrum. Should this be the case, the milled ring must be turned very slightly in the opposite direction until the horizontal lines disappear.

The rain-bands are like the Fraunhofer's lines, at right angles to the spectrum; but instead of being sharply defined, they are simply shaded bands, which become darker as rain becomes imminent. When they are very prominent, they are more conspicuous than any of the Fraunhofer lines.

The diagram by Mr. Rand Capron shows the rain-bands in the spectrum of six different degrees of intensity. In looking for the rain-bands, direct the instrument towards a bright part of the sky, at about an angle of 20° (twenty degrees) above the horizon. Mr. Rand Capron, F.R.A.S., in his well-known pamphlet, "A Plea for the Rain-band," gives the following instructions for observing with the Spectroscope:—

"For ordinary meteorological purposes, and to observe the general character of the band near D, a pocket or miniature instrument is quite sufficient—indeed, *best adapted*. This may be obtained from any scientific optician, and will cost, according to construction, from one to three guineas. The more expensive

ones have an adjustable slit and achromatic lenses ; but these are not necessities, and the cheaper form is nearly as useful.

"Having obtained the Spectroscope, close the slit and adjust the focus till the lines in the spectrum are sharp and clear. This should be done on a bright part of the sky. Then point the instrument to the quarter of the heavens which it is desired to examine, and note results as to, especially, lines D and their neighbourhood.

"I generally observe thus at 9 a.m. daily from my laboratory window (looking towards the south), but if time and opportunity allow three observations, at 9 a.m., 1 p.m., and 5 p.m. would be better, varying the parts of the sky tested ; and I examine with the Spectroscope elevated about 13 degrees. Professor Smyth recommends to point as low as you can to the horizon, provided you get transmitted light, and to observe when the sun is neither high nor low. I find in practice 9 a.m. (the same hour when my other instruments are observed) a good time to make the observation when only a single one is taken daily, and also that if I get *too* low on the horizon I am apt to have always a 'rain-band,' or rather a false band due to earth moisture. In observing you will soon remark changes in the characters of some of the spectrum lines as compared with these when seen on a blue sky with an elevated Spectroscope, and, moreover, bands of varying intensity are found added to the low spectrum not seen in the higher one. The lines and bands that change their character, or are variable in their appearing, are telluric ; either rain-bands or lines, called by Professor Smyth 'a function of moisture and temperature,' or low sun-bands and lines, distinguished by him as 'a function of dry air and low sun.' The true solar lines remain unchanged. Professor Smyth, in the Edinburgh volume before referred to, fully describes both in type and by illustrative drawings all these bands and lines, and their changes, and points out that there are several smaller rain-bands besides that near D. In practice it will, however, be found sufficient for meteorological observations generally to examine the principal one on the red, or, if you have the spectrum as I do (with the red end of the spectrum to my left hand), left side of D. An examination of the other lines, though desirable for special purposes, will only tend to confuse the general observer. Professor Smyth recommends, and has used, a dry air band on the right-hand side of D as a standard of comparison with the rain-band ; but I have not often myself made use of it, judging independently by the rain-band itself.

"In enumeration of the darkness of the band, for the purpose of record, I use from No. 1 to 5 as under :—

1 means faint.	4 means moderate to strong.
2 „ faint to moderate.	5 „ strong.
3 „ moderate.	

1 to 10 is the enumeration recommended and employed by Professor Smyth. I found, however, the dividing into so many degrees was difficult, especially when the intensity is slightly changing by passing clouds. Simultaneous observations should, of course, be made and recorded of barometer, wet and dry bulbs, and wind (force and direction); and the circumstances of sun, sky, and cloud at the time should be shortly noted.

"Ozone test papers may also be usefully referred to, as cold winds alike affect them and rain-band readings. The D lines are generally more or less involved in the rain-band shading.

"To enable the observer to judge of the general appearance and intensity of the larger rain-band near D, I have given (see *frontispiece*) some drawings of spectra as seen in a Spectroscope of small dispersion, of which the following is a description :—

"(1) Spectrum as seen upon a pure high sky, showing principal solar and telluric lines in their proper positions, and with their designations, but not showing the finer lines between, nor any bands.

"(2) Spectrum observed January 17th, 1881, 8 a.m. Morning dull; red sunrise; low sun-bands and lines (note especially band to right of D) strong. *No* rain-band.

"(3) Spectrum observed 24th August, 1881, 8 a.m., showing moderate low sun-bands and lines, and a *faint* rain-band, with rain-lines showing through.

"(4) Spectrum seen November 16th, 1880, 1 p.m. Rain and wind, but clearing in some parts of the sky. Low sun-bands and lines weak. Rain-band *moderate*.

"(5) Spectrum seen December 9th, 1880, 8 a.m. Sun shining through watery clouds; low sun-lines strong. Rain-band *strong*.

"(6) Spectrum seen July 6th, 1881. Rain-band everywhere, and *exceptionally* strong, stretching nearly half way between C and D. Whole spectrum darkened and obscured.

"The above-described drawings do not give, except in a rough way, the details of the lines and bands other than the rain-band, which is situate to the left of the double line D, and has its place marked by R.

NOTE.—Observe *only* the dark band marked Rain-band in the coloured frontispiece. This can always be found by holding a lighted lucifer match in front of the Spectroscope, when a bright yellow line will be seen just where the Rain-band should be looked for.—J. B.

"Spectra 2, 3, 4, and 5 are meant for observing by, as representing—2, none; 3, faint; 4, moderate; and 5, strong. Rain-band faint to moderate, and moderate to strong (the intermediate intensities), can be estimated, and in practice the eye and judgment will soon be found to accommodate themselves to the graduations 1 to 5."*

A valued correspondent in the North-west of Scotland writes to me:—

"I am feeling more and more confidence from day to day in the Rain-band Spectroscope. The Aneroid Barometer has been falling for days; the sky has been dull and gloomy, looking exactly like wet; but there has been no rain-band and no rain. When I left home this afternoon, at 4 p.m., the rain-band was just appearing, nebulous and hazy, and I expect there will be rain to-night."

My correspondent wrote me by the next post that rain fell smartly within a few hours.

Mr. F. W. Cory, F.M.S., of Buckhurst Hill, Essex, who has made rain-band observations for three months, writes as follows to *Knowledge*:—

"1. Adjust the focus and slit of the Spectroscope so that the lines in the spectrum may be of the clearest definition, and whilst taking an observation, shade the eyes with the hands, in order that all extraneous light may be excluded as much as possible.

"2. Observe in a general way from ten to twenty degrees above the horizon, and towards the quarter from which the wind is blowing; but if the latter be not practicable, point the instrument in a northerly, in preference to a southerly, direction.

"3. If 80 per cent. of the dark band be shown at the zenith, heavy rain will certainly occur before long.

"4. During wet weather, the amount of rain-band may be low; this for the most part denotes fine weather to follow.

"5. If the instrument be directed towards the point from which the wind is blowing, and should the clouds be passing in one and the same direction, and 20 per cent., or less, of the rain-band be shown, no rain will follow for at least six hours, in spite of any threatening appearance of the sky.

"6. With a knowledge of the course and peculiar characteristic of storms, which is not difficult to acquire (*vide* 'Aids to the Study and Forecast of Weather,' by W. Clement Sey, M.A.), it is possible to foretell rain by the help of the Spectroscope with surprising accuracy.

* Mr. Rand Capron's Pamphlet, "A Plea for the Rain-band," will be sent to any address for four stamps.

"Observers will not be disappointed if the above suggestions are carefully attended to. The principal difficulty lies in determining the percentage of rain-band ; but this is to be overcome by practice and experience, and they will find that after a time the indications of the Spectroscope can be read at a glance of a few seconds in duration."

HOW TO USE A DIRECT-VISION SPECTROSCOPE.

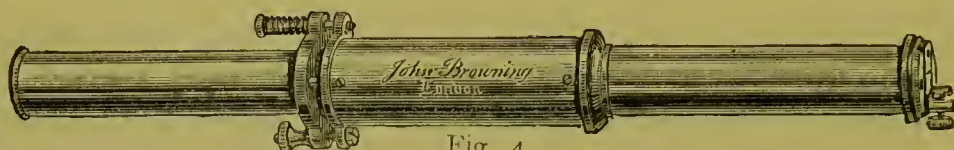


Fig. 4.

From their simplicity of construction, these instruments are the easiest to use, and are therefore the best adapted for beginners. In using them, it is only necessary to direct the slit to the source of light, which should not be a bright coal-gas or lamp flame, as they will not give any lines. A tallow candle, with a long snuff, will give the yellow or orange sodium line. Coal-gas, burnt in a Bunsen's burner, will give carbon lines. A small quantity of a salt of an alkali or alkaline earth fused on a wire and held in the flame of the Bunsen's burner, will give bright lines. (See "How to Obtain the Bright Lines given by any Substance.") Having by either of these methods obtained some bright lines, focus them carefully by moving the sliding drawer-tube of the telescope, and then close the jaws of the slit until the lines appear fine without becoming indistinct.

Any form of stand, with a ring, tube, or a clip to hold the body of the Spectroscope, provided with horizontal, or, better still, with horizontal and vertical motions, greatly facilitates the use of the instrument.

These instructions will apply to the Miniature Spectroscope.

BROWNING'S NEW MINIATURE SPECTROSCOPE, WITH MICROMETER.



Fig. 3.

This portable and complete instrument may be used for showing any of the leading experiments in spectrum analysis ; the Fraunhofer lines ; the lines in the spectra of the metals, and

the alkaline earths and alkalies ; the spectra of gases ; and absorption bands.

Applied to a telescope, it may be used for viewing the lines of the solar prominences. It can also be used as a Micro-Spectroscope. The position of the bands in any spectrum may be seen at a glance, as a photographed Micrometer scale is reflected, by means of a magnifying prism, into the field of view, so that it appears parallel with the spectrum. Each tenth line on the scale has a figure above it. This instrument is very convenient for taking the position of a line rapidly.

THE STUDENT'S SPECTROSCOPE.

This instrument has a prism of extremely dense glass of superior workmanship. The circle is divided, and reads with a vernier, thus dispensing with an illuminated scale ; this arrangement possesses the very great advantage of giving angular measures in place of a perfectly arbitrary scale.

The slit is also furnished with a reflecting prism, by means of which two spectra can be shown in the field of view at the same time.

The instrument is so arranged that, with a slight alteration of the adjustments, it can be used for taking the refractive and dispersive powers of solids or liquids. For information on this point see separate heading.

A photographed Micrometer can be applied to this instrument.

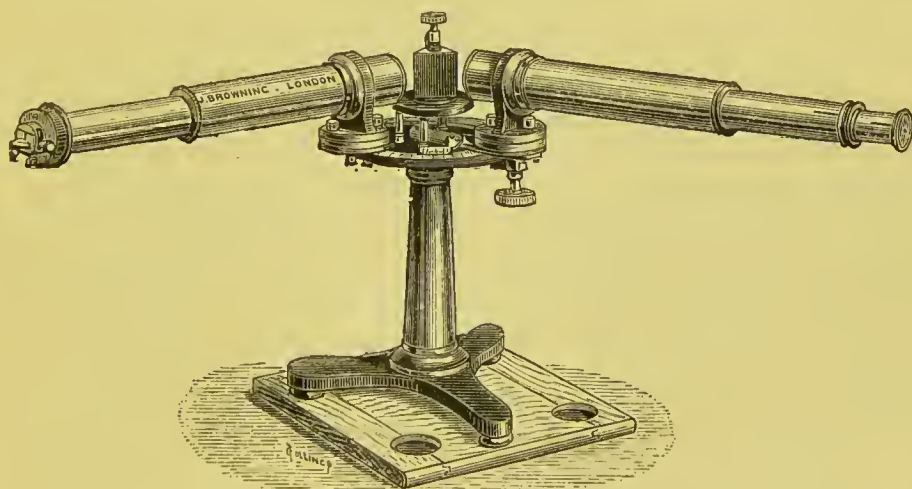


Fig. 5.

The instructions for using the Spectroscopes with two to five prisms, which follow, will also apply to this instrument.

THE MODEL SPECTROSCOPE.

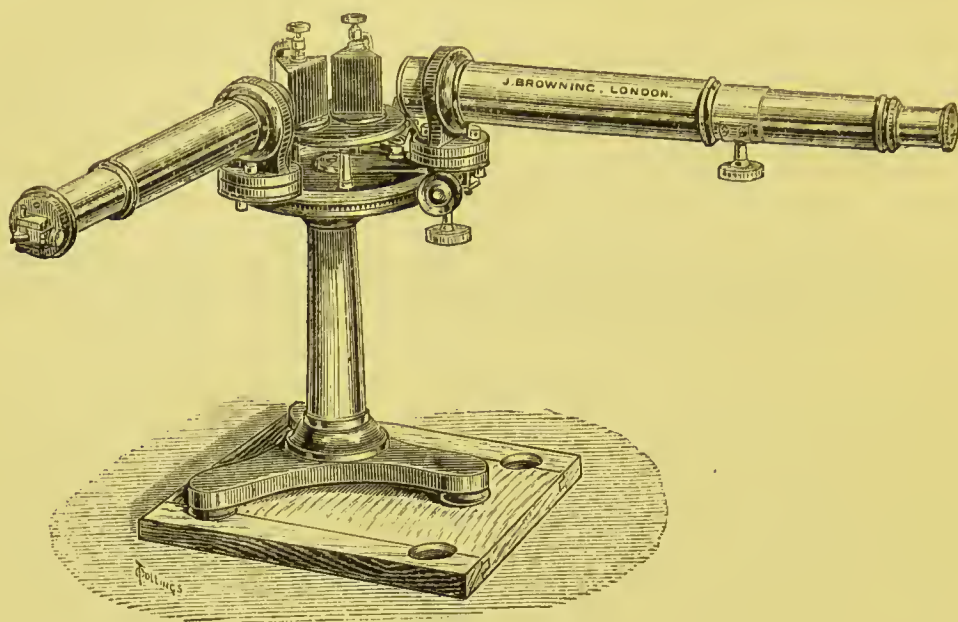


Fig. 6.

This instrument has two dense glass prisms, two eye-pieces, rack motion to telescope, and tangent screw motion to vernier. It will widely separate the D lines.

A photographed Micrometer can be applied to this instrument.

HOW TO USE THE STUDENT'S, THE MODEL, OR THE CHEMICAL SPECTROSCOPE.

Screw the telescope carrying the knife edges at the small end into the upright ring fixed on to the divided circle, and the other telescope into the ring attached to the movable index. Now place any common bright light exactly in front of the knife edges, and while looking through the telescope on the movable index (having first unscrewed the clamping screw under the circle), turn the telescope with the index round the circle until a bright and continuous spectrum is visible.

HOW TO OBTAIN THE BRIGHT LINES IN THE SPECTRUM GIVEN BY ANY SUBSTANCE.

Remove the bright flame from the front of the knife edges and substitute in its place the flame of a common spirit-lamp, or, still better, a gas jet known as a Bunsen's Burner. In the Burner shown in the engraving, there is a ring at the bottom of

the tube, with four apertures; by turning these, so that the holes are closed, the Burner gives a white light, well adapted for adjusting the instrument, and showing a continuous bright spectrum in the field of view. When this has been done, turning the ring so as to make the apertures correspond, will admit a quantity of air, and give a dull bluish, very hot flame, well adapted for spectrum analysis. Having obtained this hot flame, take a piece of platinum wire, about the substance of a fine sewing needle; bend the end into a small loop about the eighth of an inch in diameter; fuse a small bead of the substance or salt to be experimented on into the loop of the platinum wire, and attaching it to any sort of light stand or support (as Fig. 28), bring the bead into the front edge of the flame, a little below the level of the knife edges. If the flame be opposite the knife

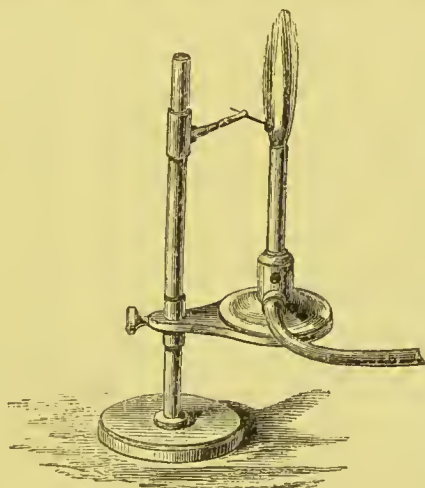


Fig. 28.

Improved Spectroscope Lamp, containing Burner and Clip, on a Single Stand.

edges on looking through the eye-piece of the telescope, the fixed lines due to the substance will be plainly visible. When minute quantities have to be examined, the substance should be dissolved, and a drop of the solution, instead of a solid bead, be used on the platinum wire. The chlorides of the alkaline earths are the best adapted for giving the spectra—as chloride of calcium, chloride of barium, &c.; but carbonate of soda and ferrocyanide of potassium give better results than the chlorides of these alkalis.

Whatever salts are used, they must be *pure*, as the salts of commerce contain many impurities, and the spectra of these impurities are sufficiently bright to mask the spectra of the substance themselves.

Where coal-gas is not obtainable, a small flame of hydrogen may be used. This should be made in an apparatus in which it is generated only while it is being consumed. The flame of a spirit-lamp may be used, but the results are unsatisfactory. The oxy-hydrogen blowpipe is the best of all the sources of heat known for this purpose; but when the bright lines of the metals are required, the Induction Coil must be used as described under a separate heading.

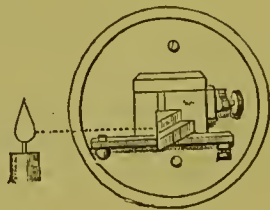
When it is desired to show the spectrum of a salt for any length of time, small pieces of pumice-stone may be soaked in a saturated solution of the salt, and then attached to a fine platinum wire, and held in the flame as described for a bead.

A frequent source of failure is using a platinum wire too thick for the purpose. The wire should be only just thick enough to support the substance without a tendency to vibration.

The delicacy of this method of analysis is very great. Swan found in 1857 (Ed. Phill. Trans., vol. xxi., p. 411) that the lines of sodium are visible when a quantity of solution is employed which does not contain more than $\frac{1}{250000}$ of a grain of sodium.

To view Fraunhofer's lines in the solar spectrum, it is only necessary to turn the knife edges towards a white cloud, and make the slit formed by the knife edges very narrow by turning the screw at the side of them. In every instance the focus of the telescope must be adjusted in the ordinary way, by sliding the draw-tube until it suits the observer's sight, and distinct vision is obtained.

It should be noted that lines at various parts of the spectrum require a different adjustment in focussing the telescope.



The small prism, turning on a joint in front of the knife edges, is for the purpose of showing two spectra in the field of view at the same time. To do this, it must be brought close to the front of the knife edges. Then one flame must be placed in the position in which the flame of the candle is shown in the

small diagram figure, and the other directly in front of the slit. On looking through the telescope as before described, the spectra due to the two substances will be seen one above the other. A known substance should be burnt in the flame, to give one of these spectra, and a material suspected of containing the substance in the other, to see if the lines in the two spectra coincide. This is termed working by comparison, and is the best method of employing the instrument for analytical purposes.

When the slit is turned towards a bright cloud, and a light is used in the position of the candle flame, the spectrum of any substance may be seen, compared with the solar spectrum. In this manner Kirchhoff determined in the solar spectrum the presence of the lines of the greater number of the elements which are believed to exist in the sun. The absorption bands in spectra may be most conveniently examined, and accurately investigated, by means of the SORBY-BROWNING Micro-Spectroscope.

TO TAKE THE REFRACTIVE INDEX OR DISPERSIVE POWER OF ANY SUBSTANCE OR LIQUID WITH A SPECTROSCOPE.

A Model or Student's Spectroscope with one or two prisms should be used for this purpose. First remove the prism or prisms from the Spectroscope; bring the telescope in a direct line with the collimator; place a positive eye-piece in the Spectroscope, having cross-wires in the field of view; make these cross-wires bisect the slit. The substance of which the refractive index is to be determined should be cut into the form of a prism of 60° . Having noted the reading of the telescope on the arc of the Spectroscope, which for this purpose should be divided to degrees and minutes, place the prism and the substance on the plate of the Spectroscope, at the minimum angle of deviation—that is, at the angle at which the ray from the collimator is deflected the least. If the solar spectrum be now observed through the telescope and prism, the position of the principal Fraunhofer lines may be read off on the divided arc by bringing them successively to coincide with the cross-wires in the eye-piece. The method of calculating the index of refraction and the dispersive power from these observations, will be found very clearly stated in Ganot's "Physics."

When a liquid has to be employed instead of a solid, a hollow prism must be used for containing it. This prism should have glass sides, which can be easily removed for the purpose

of cleansing them after each experiment. When in use, the prism and the sides are placed together in a metal frame; a screw in one side of this frame secures them in position, and confines the liquid.

Glass prisms can be made so accurately that the sides, when placed against the prism, will be kept in position by atmospheric pressure; but they are seldom employed on account of their great cost.

BROWNING'S UNIVERSAL AUTOMATIC SPECTROSCOPE.

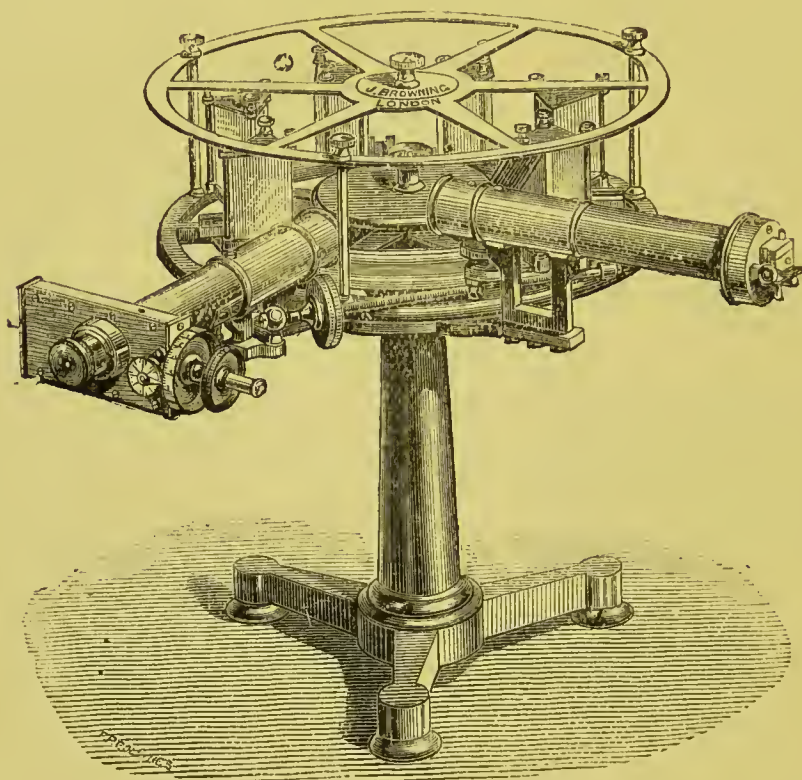


Fig. 7.

In this instrument the prisms are automatically adjusted to the minimum angle of deviation for the particular ray under examination; the position of the lines in the spectra is changed by means of a screw; the revolution of this screw adjusts the prisms automatically for the ray under measurement. The instrument has six prisms, and by means of the reversion of the ray a dispersive power of twelve prisms is obtained. By changing the position of one of the prisms, any dispersive power, from two to twelve prisms, can be used at

pleasure, without deranging any of the adjustments of the instrument. This Spectroscope is therefore applicable to every class of spectrum work either in the laboratory or observatory.

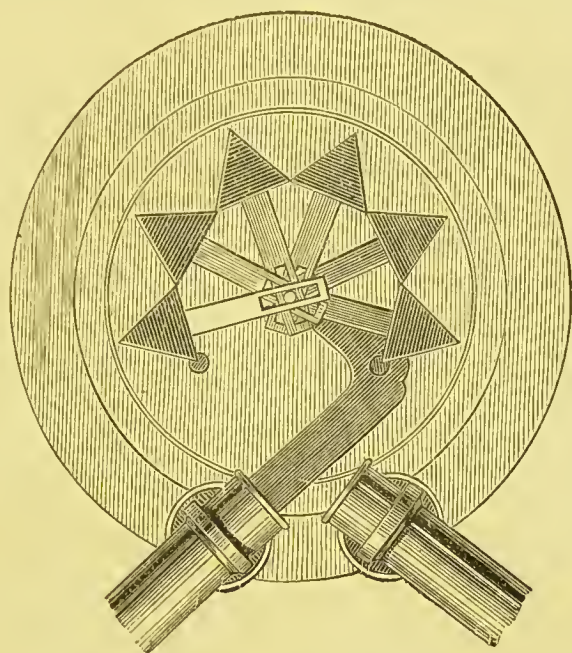


Fig. 8.

Diagram showing the Automatic Action by which the Prisms are automatically adjusted to the minimum angle of deviation for the particular Ray under examination.

In an ordinary Spectroscope the prisms are usually adjusted to the minimum angle of deviation for the most luminous rays in the spectrum—by preference, I adjust them myself for the ray E in the solar spectrum. This being done, the prisms are screwed, or otherwise firmly clamped, to the main plate of the Spectroscope. Thus adjusted, they are liable to two sources of error, one of which places the observer at a serious disadvantage. First, only the particular ray for which the prisms have been adjusted is seen under the most favourable circumstances, for only this ray passes, as all should do, through the train of prisms parallel to the base of each prism. Of more importance than this, however, is the fact that the last prism of the train being fixed while the telescope through which the spectrum is viewed is movable around an arc, it is only when the central portion of the spectrum is being examined that the whole field of the object-glass is filled.

Bunsen and Kirchoff, when making their celebrated map

of the solar spectrum, adjusted the prisms they used (four in number) for each of the principal Fraunhofer lines; but the trouble of doing this is so great that few observers have ever seen the extreme portions of the solar spectrum under favourable circumstances.

The diagram (Fig. 8) shows the method in which the change in the adjustment of the prisms to the minimum angle of deviation for each particular ray can be made automatically. In this diagram there are six prisms. All these prisms, with the exception of the first, are unattached to the plate on which they stand. The first prism is attached by one corner to the plate by a pivot on which it turns. The triangular plates on which the prisms rest are hinged together at the angles corresponding to those at the bases of the prisms. To each of these bases is attached a bar, perpendicular to the base of the prism. As all these bars are slotted and run on a common centre, the prisms are brought into a circle. This central pivot is attached to a dovetailed slide about two inches in length, placed on the under side of the main plate of the Spectroscope, which is slotted to allow it to pass through. On moving the central pivot, the whole of the prisms are moved, each to a different amount in proportion to its distance in the train from the first or fixed prism, on which the light from the slit falls after passing through the collimator, C. Thus, supposing the first prism of the train of C, represented in the diagram, to be stationary, and the second prism to have been moved through 1° by this arrangement, then the third prism will have been moved through 2° , the fourth through 3° , the fifth through 4° , and the sixth through 5° .

A screw gives motion to a lever which is attached to the last prism of the train. By turning this screw until any particular portion of the spectrum appears in the field of view, the rays which issue from the centre of the last prism are made to fall perpendicularly upon the centre of the object-glass of the telescope, T, and thus the ray of light under examination travels parallel to the bases of the several prisms, and ultimately along the optical axis of the telescope itself, and thereby the whole field of the object-glass is filled with light.

Thus the apparatus is so arranged that, on turning the screw so as to make a line in the spectrum coincide with the cross-wires in the eye-piece of the telescope, the lever L, attached to the prisms, sets the whole of the prisms in motion, and adjusts them to the minimum angle of deviation for that portion of the spectrum.

Fig. 8*.

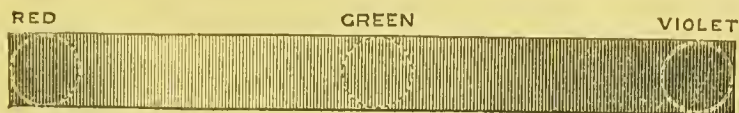


Fig. 8**.

Diagrams 8* and 8** represent the appearances presented looking through the telescope from which the glasses have been removed. In diagram 8* it will be seen that the whole circle of the object-glass is filled with light, as I have just described is the case with the new arrangement ; while diagram 8** shows the effect of moving the telescope through the angle in front of the fixed prism.

Here, in the red and violet, *where the light in the spectrum is faintest*, only about one quarter of the field of view is illuminated.

HOW TO USE AN AUTOMATIC SPECTROSCOPE. †

Although this Spectroscope is the most powerful, and in appearance the most complex made, it is very easy to use.

To vary the dispersive power of the instrument, let us suppose that the movable right-angled reflecting prism, which reverses the ray, is at the end of the train—that is, the farthest from the collimator. Remove any prism of the battery from its place in the train by sliding it with its holder out of the groove. Take out the reflecting prism in the same way, and slide it into the groove previously occupied by the reflecting prism. In this way the dispersive power of one prism, or of the whole train of prisms, may be employed at pleasure.

The subsequent manipulation will be the same as described under the heading, "How to Use a Chemical Spectroscope," with the exception of mapping the spectra. The method of doing this I have described under the heading, "How to Map a Spectrum with the Automatic Spectroscope."

p. 14

BROWNING'S AUTOMATIC SOLAR SPECTROSCOPE.

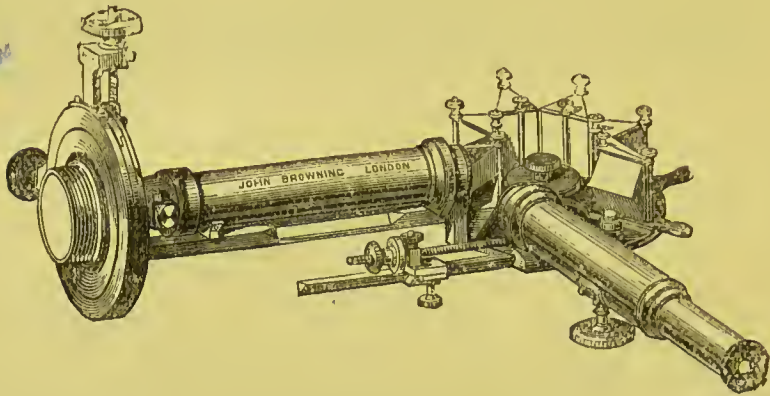


Fig. 9.

Dr. Henry Draper's important discovery of the presence of oxygen in the sun, described in *Nature*, No. 409, August 30, 1877, will direct renewed attention to the solar spectrum.

The Automatic Solar Spectroscope, figured above, will show the solar spectrum with exquisite definition, and if attached to

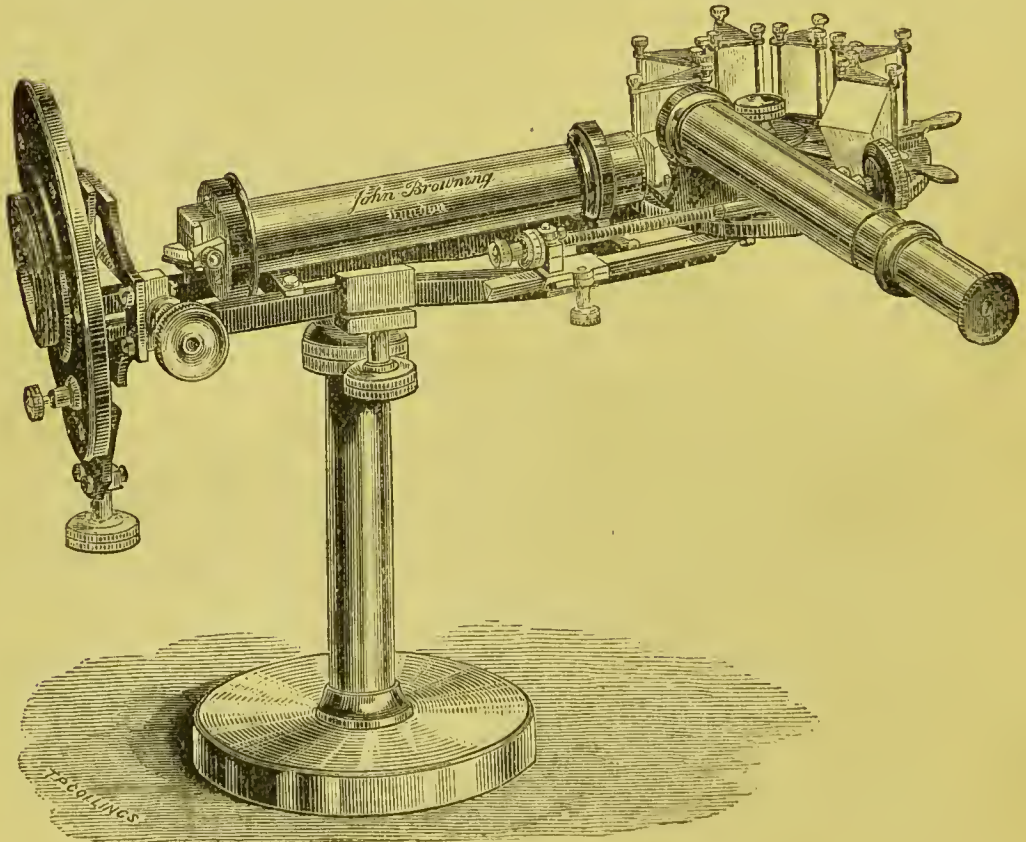


Fig. 10.

the eye-piece of a telescope of three inches or more in diameter, it will show the forms of the solar prominences.

As this Spectroscope can be used with any dispersive power from two to ten prisms, it can be arranged for observing the spectra of the stars and nebulæ. Without a telescope it can be employed for any kind of work in spectrum analysis.

By means of the reversion of the ray this Spectroscope gives a dispersive power equal to ten prisms, and this dispersive power may be changed at pleasure by the observer. The instrument is very light, and can be adapted to a telescope as small as three inches in aperture. It is provided with a movement of rotation for searching for solar prominences. *vide p 21*

The Solar Automatic Spectroscope (Fig. 10) is well adapted for use with any telescope, either a reflector or refractor, from six to twelve inches in aperture; and it can be used on a table stand, as shown in the engraving, for viewing the spectra of metals, salts, or gases.

HOW TO USE A SOLAR SPECTROSCOPE.

This instrument must be used with an astronomical telescope. No eye-pieces must be used with the telescope. The first point to attend to in using a Solar Spectroscope is to place the slit or knife edges of the instrument exactly in the focus of the object-glass or speculum of the telescope.

This must be done by projecting the image of the sun on a card without any eye-piece in the telescope. Note the distance from the end of the telescope, not the end of the sliding drawer, at which a sharp image is produced. Then attach the Spectroscope to the telescope, and move the sliding drawer until the slit is at the required distance from the end of the telescope. The best Solar Spectroscopes are so contrived that the face of the slit can be seen when they are in the telescope. In this case a card may be placed on the face of the slit, and the drawer tube of the telescope moved until the sun's image is seen sharply in focus, when the card may be removed.

Now close the jaws of the slit, and move the telescope, so that the sun's limb or *edge* of the disc falls across the slit at right angles to the direction of the slit. Turn the Spectroscope round, so that every point of the sun's disc passes over the slit while looking through the instrument. If at the time of making the observation there are any prominences on the sun, a faint red line, and possibly a blue and a green line, will also be visible on

looking into the Spectroscope. These lines will appear brighter than the other parts of the solar spectrum.

Having obtained the red line, bring it to the centre of the field of view, and carefully open the jaws of the slit while looking at the red line.

If the prominence should be of a well-marked character, its form will then be seen. The problem, when using the Spectroscope, is to get a small bright image into a very narrow slit. In consequence of their large aperture and short focal length, the new Silvered Glass Reflecting Telescopes are admirably adapted for working with the Spectroscope.

THE STAR SPECTROSCOPE.

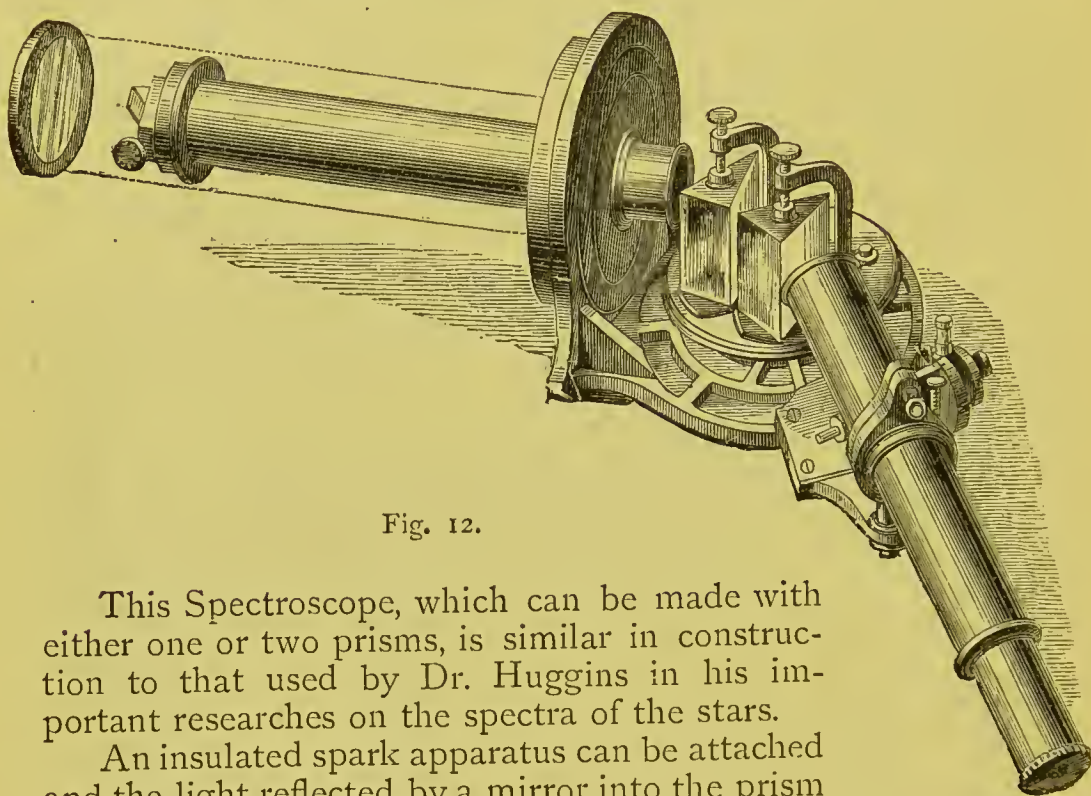


Fig. 12.

This Spectroscope, which can be made with either one or two prisms, is similar in construction to that used by Dr. Huggins in his important researches on the spectra of the stars.

An insulated spark apparatus can be attached and the light reflected by a mirror into the prism in front of the slit, for obtaining the spectra of the metals for comparison, to either of the above instruments.

THE AMATEUR'S STAR SPECTROSCOPE.

This is a Direct-Vision Spectroscope, and is very easy to use with an equatorial provided with a clock; but the new McClean's Star Spectroscope has the great advantage that it can be used on any alt-azimuth stand with the greatest facility.

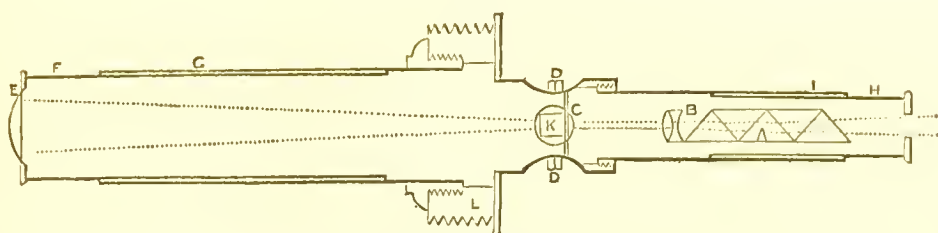


Fig. 13.

Section of Browning's Amateur's Star Spectroscope.

A is a compound direct-vision prism, consisting of five prisms. B is an achromatic lens, which focusses on the slit C, by means of a sliding tube, H; both the prisms and the lens are fastened in this tube. K is a small right-angled prism, covering half the slit, by the aid of which light may be seen reflected through the circular aperture in front of it. In this manner a comparison may be made with the spectra of metals or gases. The reflecting prism, with the ring to which it is attached, can be instantly removed, and the whole length of the slit used if desired. DD is a ring milled on the edge; on turning this round, both edges of the slit recede from each other equally, being acted on by two hollow eccentrics. The lines can thus be increased in breadth without their original centres being displaced—a point of importance. E is a cylindrical lens attached to the tube F, which slides in another tube, G. To use the Spectroscope on a telescope, the adapter needs only to have a thread which shall enable it to be screwed into the draw-tube of the telescope, in the place of the ordinary Huyghenian eye-piece. A photographed Micrometer can be added to this instrument.

The draw-tube must then be adjusted so that the slit C comes exactly to the focus of the object-glass. When stars, &c., are about to be observed, this point should be ascertained beforehand, by the aid of an image of the sun, some suitable mark to indicate the focus being made on the draw-tube of the telescope. When this has been once done, the tube can be set by this mark, and the Spectroscope screwed in at any time without any trouble in adjustment.

HOW TO USE A STAR SPECTROSCOPE.

The Spectroscope should occupy the place of an eye-piece in the telescope.

The knife edges or jaws of the slit should be exactly in the focus of the object-glass.

In this position, if the cylindrical lens is removed, the spectrum of a star will be a mere line of light.

The cylindrical lens is for the purpose of widening this line to such an extent that the lines in the spectrum may readily be discerned ; for this purpose the lens must be placed *with its axis at a right angle to the slit*, and the best distance from the slit will be between three and six inches.

The nearer it is brought to the slit the broader will be the spectrum, but it should not be used too close, on account of the diminution of the light.

When it is desired to obtain the spectra of planets, comets, or nebulae, or indeed any heavenly bodies possessing considerable diameter in the telescope, the cylindrical lens may advantageously be dispensed with.

MCCLEAN'S NEW STAR SPECTROSCOPE (PATENT).



Fig. 11.

Many persons well acquainted with the solar spectrum have yet never seen the beautiful and almost infinitely varied spectra of the stars. The Star Spectroscopes in general use are expensive and difficult to manipulate with. This arises from the fact that, in most instruments, the image of a star is required to

fall within the jaws of a narrow slit, not more than $\frac{1}{1000}$ inches in width, and an equatorially mounted telescope with clock-work is almost indispensable for using them. Star Spectroscopes of simpler construction, both with and without cylindrical lenses, have been made, but their performance has not been found satisfactory. In the instrument contrived by Mr. McClean, exquisitely fine lines can be seen in the spectra of stars without the use of any slit. The slit being dispensed with, the instrument can be used on any telescope without a clock, or even on any alt-azimuth stand; instead of a slit, a concave cylindrical lens is used, to bring the lines of the spectrum to a focus on the retina.

HOW TO USE MCCLEAN'S STAR SPECTROSCOPE.

To use this Star Spectroscope, it is only necessary to bring the star to the centre of the field of view, remove the eye-piece, and insert the Spectroscope instead. The Spectroscope is mounted in a tube, the same size as Browning's Achromatic Eye-pieces; as these slide into an adapter, the change can be effected very quickly, without shaking the instrument. Mr. McClean has adapted a revolving nose-piece to his telescope, one arm of which carries the eye-piece, and another the Spectroscope; and, by the aid of this contrivance, the change from one to the other can be made instantaneously. It is necessary to observe that, this Spectroscope having a negative lens, the lens should be inside the solar focus of the object-glass or mirror employed; roughly, the eye-cap of the Spectroscope should be placed at the solar focus. If the motion of the telescope is not kept up at a proper rate, the light of the spectrum wanes; but being without a slit, the spectrum of a star is never entirely lost, and the observer can, by giving a slight motion to the telescope, either by screws or any other means of adjustment with which the stand may be provided, give the necessary motion to the telescope, to keep the spectrum always as bright as possible.

The Spectroscope is applicable for any telescope of three inches and upwards in diameter.

By throwing the image of an illuminated point into the field of view, any spectra can be seen and used for the purpose of comparison—the required point being illuminated either by a Bunsen's burner, in which salts are being ignited, or by an induction tube, or the electric spark, taken between electrodes of various metals.

Mr. Browning has the exclusive right of making the Spectroscope.

An adjustable slit and convex lens can be used as an addition to McClean's Star Spectroscope, for showing the Fraunhofer lines in the solar spectrum, the bright lines of the metals, alkalies, gases, &c.

HOW TO USE McCLEAN'S UNIVERSAL SPECTROSCOPE.

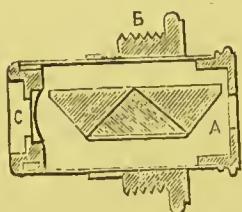


Fig. 11*.

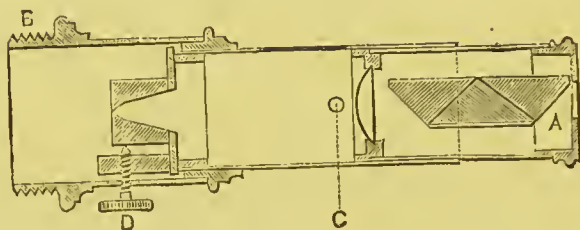


Fig. 11**.

The diagram (Fig. 11*) is a section of McClean's Star Spectroscope as it is used in an astronomical telescope for viewing the spectra of the stars.

When it is desired to use the instrument for chemical purposes, or for showing the Fraunhofer lines, it must be used without a telescope. Take out the small concave cylindrical lens C, which is fitted tightly into the end of the short tube A, containing the prism (Fig. 11*).

Insert the small convex lens, which will be found in the case, in its place.

Remove the adapter B, in which the small tube containing prisms has been used with the telescope.

Now place the tube A, containing the prisms and *convex* lens, in the tube carrying the slit, and proceed to use it as in the instructions for using a Direct-Vision Spectroscope.

The Spectroscope, as now arranged, is shown in the diagram (Fig. 11**). Note that the lines on the tube containing the prisms and the tube carrying the slit must be made to coincide while focussing the lines in the spectrum.

The small milled head D in this diagram serves to regulate the width of the slit.

The second arrangement of the Spectroscope (Fig. 11**) may be used for viewing the chemical spectra, the spectra of the metals, the spectra of gases in induction tubes, or absorption bands in liquids.

Screwed into an astronomical telescope by the coarse screw known as the astronomical thread, B (Fig. 11**), it will show the bright lines of the gaseous nebulæ, the spectra of the planets, or the bright lines of the solar prominences.

THE INDUCTION COIL,

For working Induction Tubes, giving the Spectra of the Gases, or for obtaining the Spectra of the Metals by the aid of the Electric Spark.

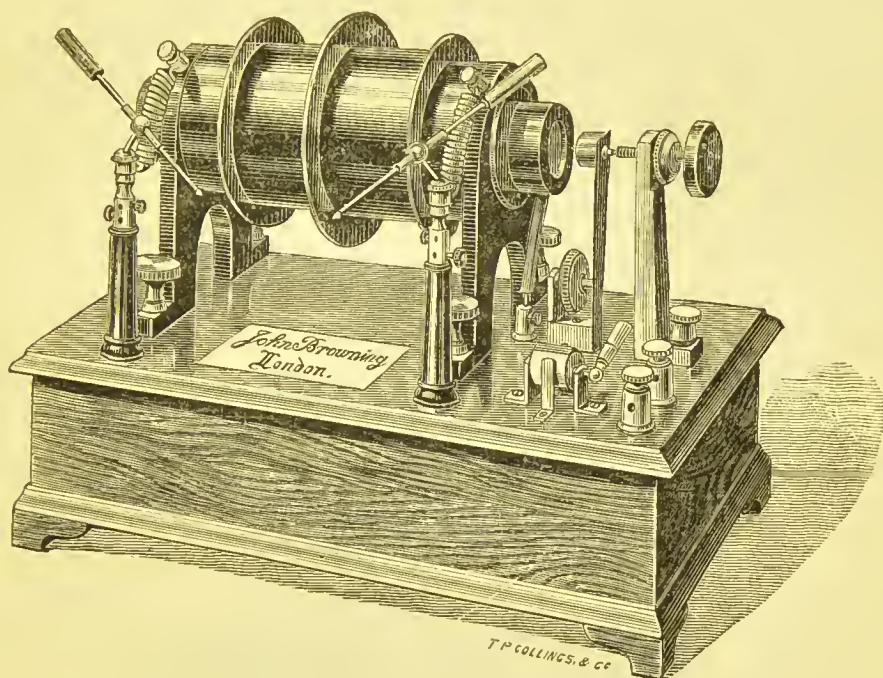


Fig. 14.

REQUIRED FOR WORKING INDUCTION TUBES.

Either an Induction Coil, to give a $\frac{1}{2}$ -in. spark in dry air, which should be used with 1 quart-size Bunsen's cell ;

An Induction Coil, to give a 1-in. spark in dry air, with 2 quart-size Bunsen's cells ; or—

An Induction Coil, to give a $1\frac{1}{2}$ -in. spark in dry air, with 3 quart-size Bunsen's cells.

REQUIRED FOR OBTAINING THE SPECTRA OF THE METALS.

Either an Induction Coil, to give a $2\frac{1}{2}$ -in. spark in dry air, with 1 quart-size Bunsen's cell ;

An Induction Coil, to give a $3\frac{1}{2}$ -in. spark in dry air, with 3 quart-size Bunsen's cells ;

An Induction Coil, to give a $4\frac{1}{2}$ -in. spark in dry air, with 5 quart-size Bunsen's cells ; or—

An Induction Coil, to give a 6-in. spark in dry air, with 6 quart-size Bunsen's cells.

Where the trouble of charging Bunsen's cells is objected to, or it is desirable to avoid the nitrous fumes they give off, bichromate batteries can be employed. These batteries are very cleanly, but not nearly so powerful as the Grove's or Bunsen's batteries, so that the Coils will not work with their full power when they are used.

A bichromate battery can be arranged so that, by using a winch, the elements may be removed from the exciting solution at pleasure. These batteries may be used many times without re-charging. One of these batteries is shown in the engraving facing page 31.

BROWNING'S SPARK CONDENSER.

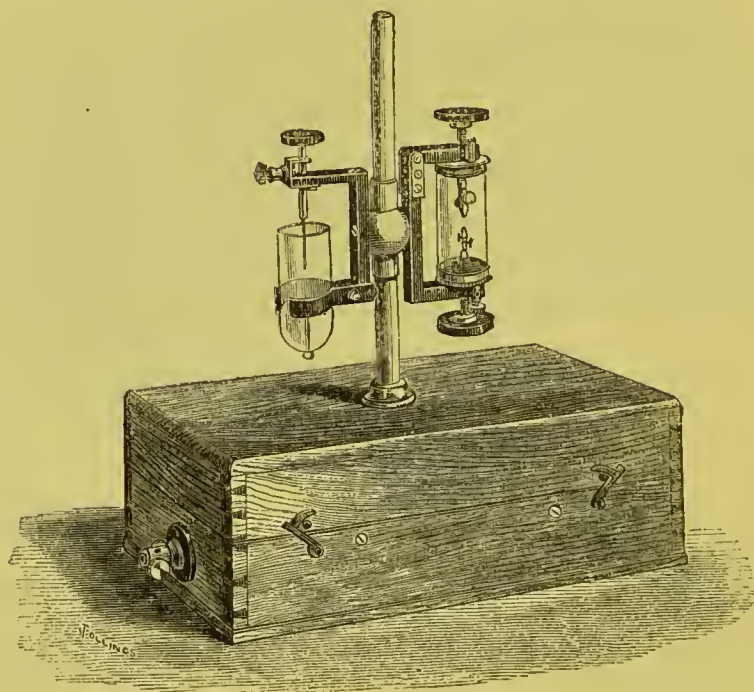


Fig. 15.

This contrivance is designed to replace the Leyden Jars which are generally used with Induction Coils to increase the temperature of the spark when it is required for spectrum analysis. The apparatus consists of an arrangement of ebonite plates coated with tinfoil, and enclosed in a mahogany case. Any amount of surface may be used at pleasure by rotating the circle on the top of the case. Unlike the Leyden Jar, the action of

the apparatus is not affected by damp. A very convenient arrangement for holding the metals of which the spectra are required, screws on to the lid of the case, and when not in use packs inside the lid.

Becquerel's apparatus for obtaining continuous spectra from solutions of salts of the metals is attached when required. It is shown on the left of the upright rod in Fig. 15.

HOW TO USE BROWNING'S SPARK CONDENSER.

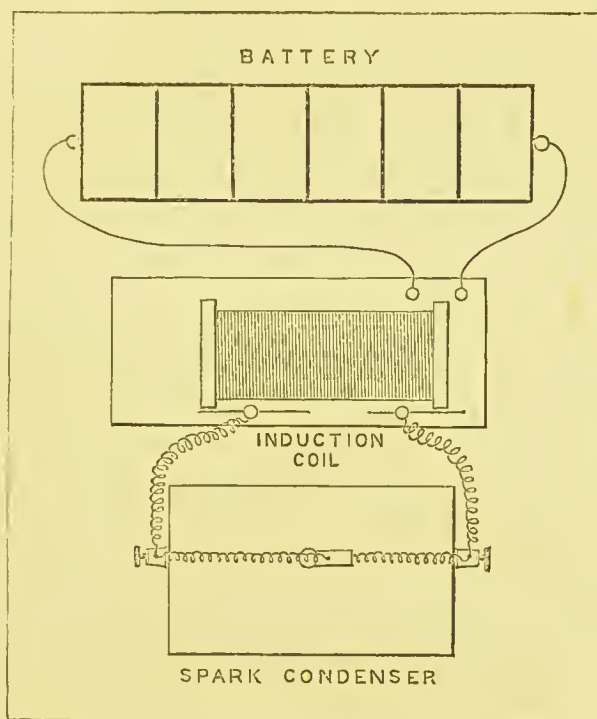


Fig. 15a.

Connect the wires from the battery with the two clamp screws at one end of the Induction Coil. Then carry a fine wire from each of the terminals of the coil (the points from which the sparks are given), one to each clamp screw at the opposite ends of the Spark Condenser. If the commutator of the coil be now turned on, the spark will pass between any pieces of metal placed in the two pairs of tweezers on the insulated ebonite support of the Spark Condenser. This spark will be very different from that of the Induction Coil—being shorter, thicker, and much more brilliant. A spark given through the Condenser, from $\frac{1}{10}$ to $\frac{1}{8}$ of an inch long, is best adapted to give the spectra of the metals in the Spectroscope.

The diagram (Fig. 15a) will perhaps show more clearly

than any description how the connections between the battery, spark condenser, and coil are to be arranged. When a Leyden Jar is used, the connections are arranged in a similar manner, but the two wires from the coil must be connected with the inside and outside coating of the Leyden Jar.

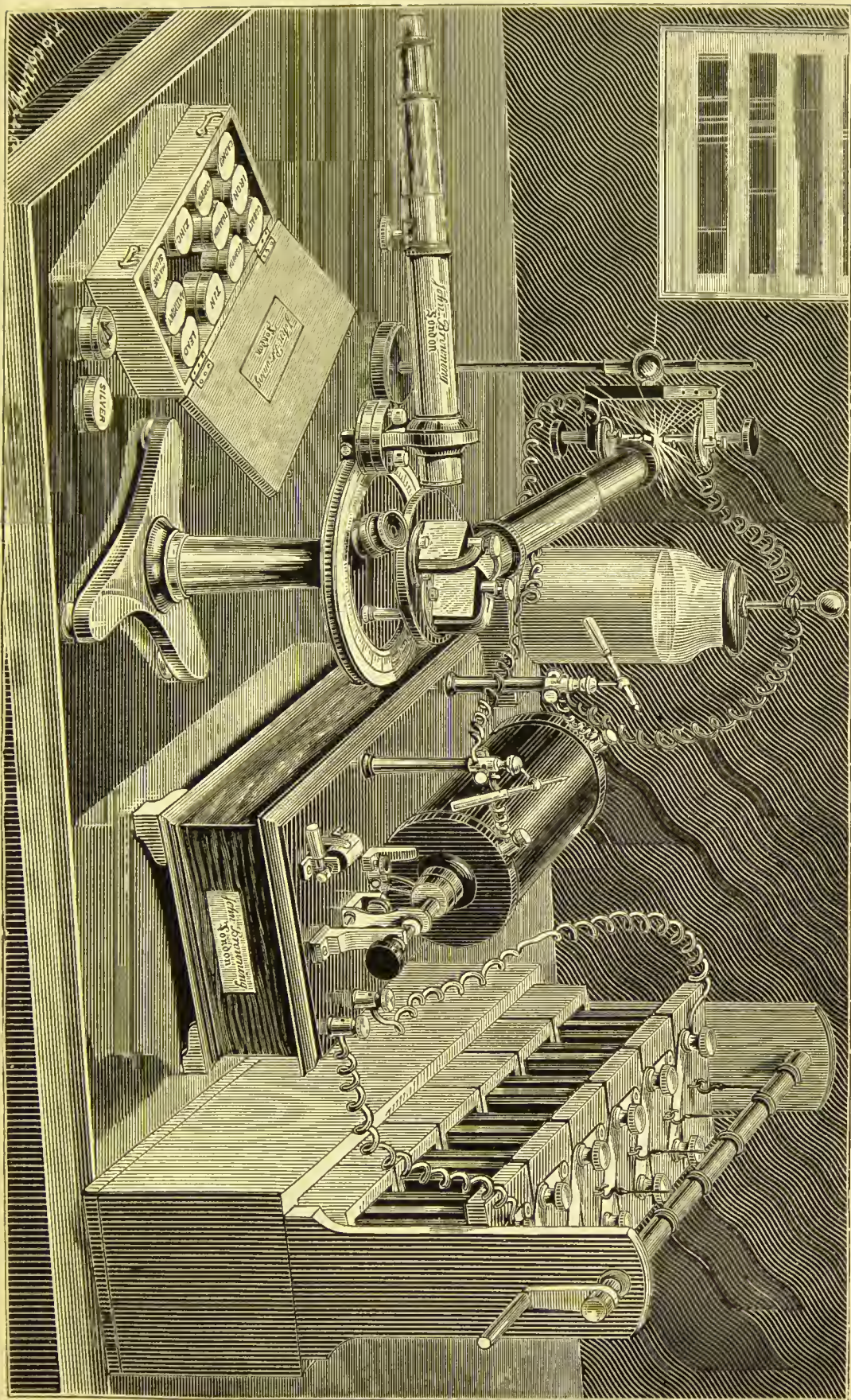
HOW TO OBTAIN THE SPECTRA OF THE METALS.

For the purpose of obtaining the spectra of the metals, use an Induction Coil. The coil should give sparks at least two inches long in dry air. Unless you have a Spark Condenser, a Leyden Jar should be introduced in the circuit, as shown in the engraving facing page 31, for the purpose of increasing the temperature of the spark. Two small pieces of the metal of which the spectrum is required should be placed in forceps attached to the terminals of the Induction Coil. These pieces of metal should be brought within one-eighth of an inch of each other. The spark should pass in a vertical line parallel to and in front of the slit.

The Leyden Jar must be connected with the Induction Coil in the following manner: Attach a wire to the metal rod which supports one pair of forceps on the terminal of the coil, and carry this to the outside covering of the Leyden Jar. A second wire should be attached in a similar manner to the other pair of forceps, and connected with the inside covering of the Leyden Jar. This suffices to bring the Jar into the circuit. When the spark from the metal is obtained the further manipulation will be the same as described, with a bead of salt on a platinum wire, under the heading of the "Chemical Spectroscope."

HOW TO USE BECQUEREL'S APPARATUS.

This Apparatus is shown attached to the Spark Condenser (Fig. 15), on the left of the upright rod. Make a concentrated solution of a salt of the metal. Pour this solution into the glass tube until it covers the platinum wire in the bottom part of the tube by about one-eighth of an inch. Then, by turning the screw with an ebonite head, bring down the upper platinum wire until it is about one-eighth of an inch above the surface of the solution. Having done this, attach one wire from the Induction Coil to the upper platinum wire, and the other to the lower platinum wire. On turning the commutator of the Induction Coil, the spark will pass through the liquid, and on bringing the slit of the Spectroscope close to the side of the tube, the spectrum of the metal which is in solution may be obtained for a considerable time.



LOCKYER'S REVOLVING SPARK APPARATUS.

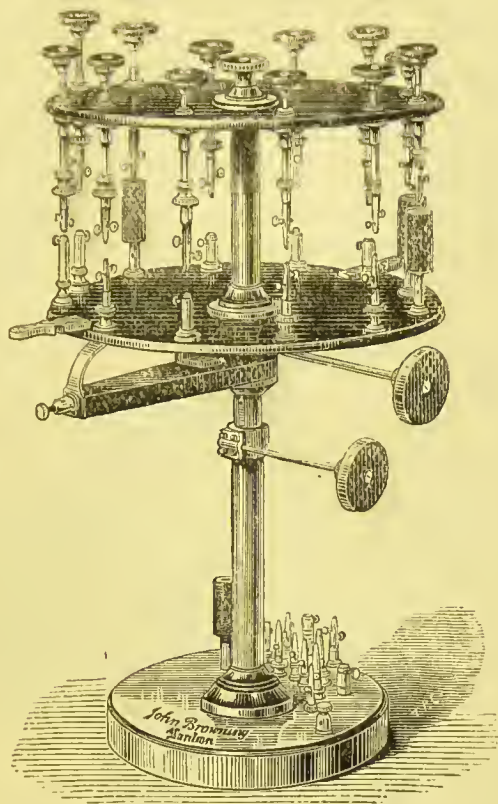


Fig 29.

This is a contrivance for the purpose of holding a number of pieces of different metals or metallic salts in small carbon crucibles. The metals are held in small pairs of tweezers, each pair being insulated. They are provided with independent adjustments. There is a vertical adjustment by means of a rack and horizontal movement given by an endless screw. The spark only passes through the tweezers in front of the Spectroscope. By means of these adjustments, the various metals can be brought in front of the slit of the Spectroscope, and their spectra obtained with the greatest ease, rapidity, and certainty.

BESSEMER SPECTROSCOPES.

Either of the Direct-Vision Spectroscopes, enumerated on pages 12 and 13, is well adapted for viewing the Bessemer flame, and great numbers are in constant use for this purpose in all parts of Europe, the highest power being best adapted for the purpose;

but the writer has devised a special instrument (the Direct-Vision Bessemer Spectroscope, with ten prisms), of very great dispersive power, having an eye-piece of large field, which shows the whole of the spectrum, giving admirable definition in all parts of the field.

The instrument below (Fig. 16) is a still more powerful instrument. The telescope has a motion between pivots, near the top of the case. There are cross-wires in the field of view to assist the observer in concentrating his own attention, or

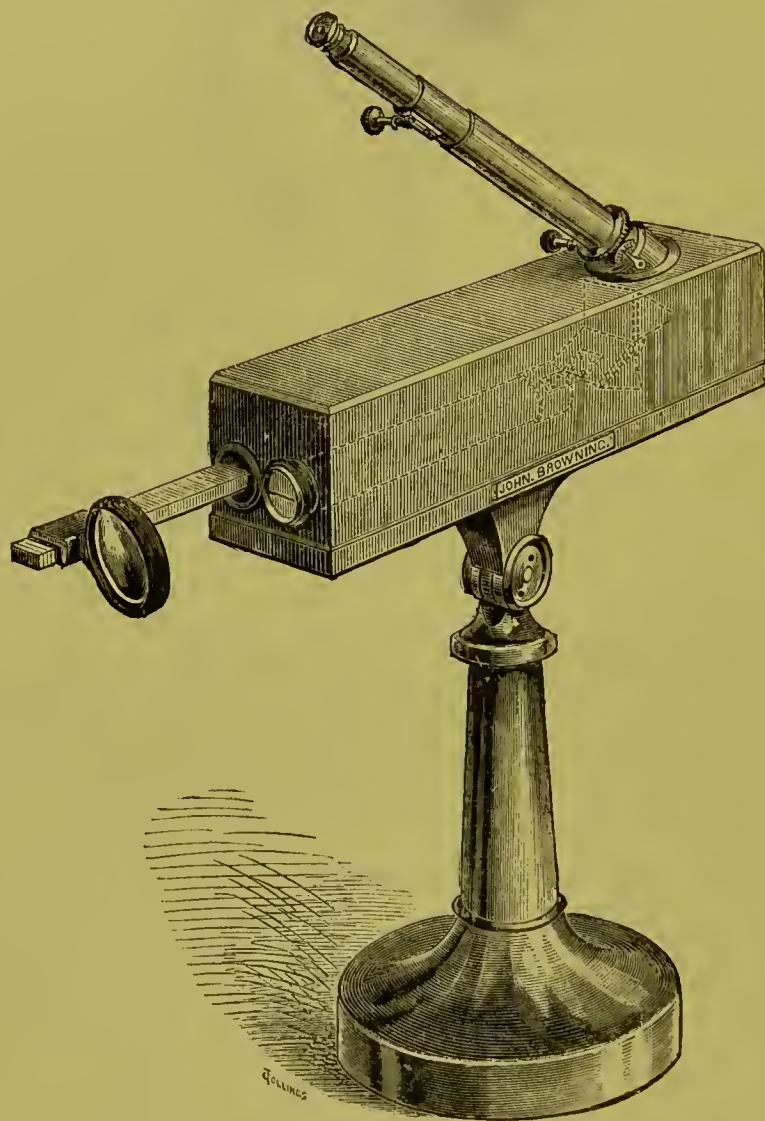


Fig. 16.

directing that of others, to any particular line in the spectrum. This instrument is so contrived that the back of the observer is

turned to the brilliant flame, which renders vision much easier. A condensing lens, shown in the engraving, which works on a rod in front of the slit, can be fixed so as to produce an image of the flame on the slit; by moving the instrument about on the hinged joint and swivel, the spectrum of any portion of the flame can be examined at pleasure. With this arrangement the Spectroscope can be used without disadvantage at any distance from the flame. The slit is protected from the action of dust by means of a glass cover, and when the instrument is not in use it can be unscrewed and enclosed in the box.

HOW TO USE A BESSEMER SPECTROSCOPE.

In the Bessemer process several tons of iron are placed in a large vessel, and when in a state of fusion, air is driven through it from apertures in the bottom of the vessel. After about twenty minutes the iron is converted into steel. During the process a flame of almost overpowering brilliancy issues from the mouth of the vessel. The conversion is almost instantaneous, and is known by a change in the colour of the flame and a reduction of its brilliancy. It requires, however, considerable practice to tell by naked eye observations when the process should be stopped; and intently watching the great glare is most trying to the sight. By the aid of the Spectroscope the completion of the process may be determined without any experience with the utmost ease and certainty.

The spectrum of the Bessemer flame is full of bright lines, a number of green lines being the most brilliant. At the instant of complete conversion these bright green lines suddenly disappear. At this moment the blowing should be stopped. Either of the Direct-Vision Spectroscopes described in this pamphlet may be used for the Bessemer process, but those specially made for the purpose, and termed Bessemer Spectroscopes, will give the most certain results. The engraving (Fig. 16) shows the instrument devised by the writer for Sir John Brown's Steel Works, Sheffield. When using this instrument the observer's back is turned to the bright Bessemer flame. The whole instrument, except the telescope, is enclosed at all times in its mahogany case, and the slit is protected also from the dust caused by the blowing process by a disc of glass.

There is a pointer in the field of view of the eye-piece; this may be set to any particular line in the spectrum, and the workman may be instructed to stop the process when this disappears.

For full information respecting the spectra given by the Bessemer process, see Roscoe's "Spectrum Analysis."

THE SMALL AUTOMATIC ELECTRIC LAMP.

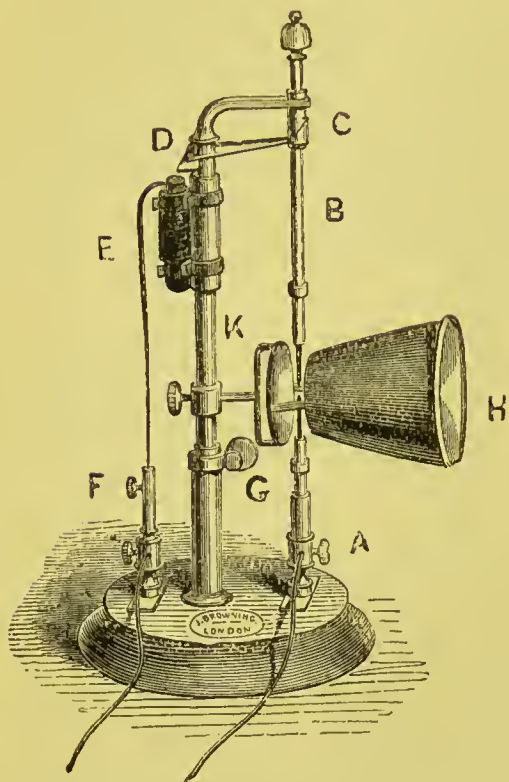


Fig. 17.

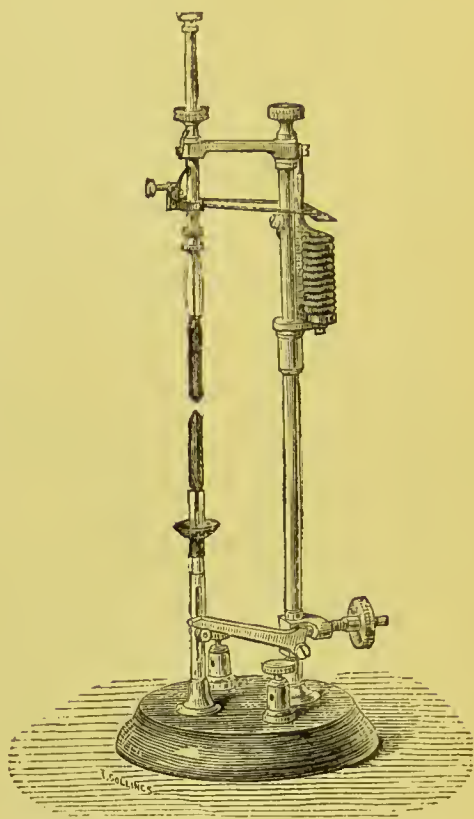


Fig. 18.

In the engraving (Fig. 17) the carbon points are carried by the holders, A B, which are provided with rings like a porte-crayon, to clamp the points when in position. C D is a soft iron feeder ; the end, C, of this feeder is so arranged that a very slight pressure on the feeder clamps the rod B, and prevents it from descending. E is a rod of soft iron, in the form of a horse-shoe ; when the electricity passes through the wire wound upon this horse-shoe, the iron becomes a magnet, and attracts the feeder. F and G are clamping screws, to clamp the sliding rods in any required position. H is a silvered parabolic reflector, for throwing the light of the Lamp to a great distance.

The small Automatic Electric Lamp (Fig. 18) is of similar construction, and can be set in action in the same manner ; but it has an additional movement for regulating the height of the carbon poles when burning. This action is influenced by the small milled head on the right-hand pillar, near the base of the stand. Should the poles burn unequally, a few revolutions of

this screw will keep them burning always at the same height. This motion is also especially useful when the Lamp is used for showing spectra in screen experiments. In spectrum analysis it should be used in the following manner : The small milled head resting on the top of the arm at right angles to the upright pillar should be turned round, until a spring falls into the oval opening ; it then catches the vertical rod and holds the upper carbon fast. A hollow having been made in the lower carbon, and the metal of which the spectrum is desired having been placed in the small crucible thus formed, the lower carbon is raised by the action of the small screw before referred to, on the right-hand pillar ; when it has been brought into contact with the upper rod, it is carefully separated until the best result is obtained on the screen. More detailed information on this subject will be found under the heading, "How to Show Spectra on a Screen."

The sharpness of the spectrum will be influenced by the width of the slit on the nozzle of the Lantern. The closer the slit the purer will be the colours of the spectrum and the better the definition of the bright lines or absorption bands when liquids are used. Of course the limit is quickly reached at which the slit must be left, or otherwise the results would become unsatisfactory from want of sufficient light.

HOW TO USE THE SMALL AUTOMATIC LAMP.

Release the clamps, F G ; place two pieces of fine hard carbon in the holders ; the carbons should be well pointed ; wipe the rod, B, with a leather, so that it may slide freely ; then adjust the large central rod so that the extreme point of the upper carbon exactly rests upon the lower carbon. Attach the wire from the last plate of zinc in the battery to the lower carbon holder, and the wire from the plate of platinum at the opposite end of the battery to the upper carbon holder. If the light should not burn steadily, alter the position of the magnet by means of a small set screw between the ends ; this screw is not shown in the drawing. The magnet must not be put close to the feeder ; the best distance to place the magnet from the feeder is generally about half an inch, but this will vary with the power of the battery employed.

The Lamp is regulated by means of a small screw, shown in the diagram at C (Fig. 17). This must be done while the battery is attached to the Lamp. By a few trials, turning this screw

backwards and forwards, it will be found that the light of the Lamp will become continuous, and it may be left even half an hour at a time, giving a steady light without any attention. Twenty quart cells are quite sufficient to work the Lamp well ; if more are used it becomes too hot. When correctly adjusted there should be no perceptible motion of the feeder.

BROWNING'S LARGE ELECTRIC LAMPS OR REGULATORS.

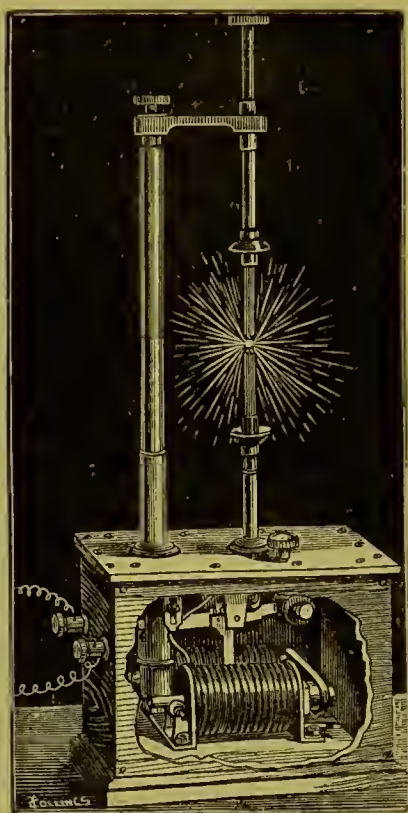


Fig. 19.

In these Regulators both carbons are moved by the electricity of the battery employed (without the aid of clockwork) ; the light remains uniform in height and more steady in action than any of the expensive Regulators previously introduced.

The medium-size Automatic Regulator (Fig. 19).—This Lamp works well with from 20 to 30 pint Grove's cells, or the same number of quart Bunsen's.

The large-size Automatic Regulator.—From 25 to 50 quart Grove's cells, or the same number of two-quart Bunsen's, should be used with this Lamp.

HOW TO USE BROWNING'S LARGE AUTOMATIC LAMP.

As the Electric Lamp is an essential part of Screen Spectrum Apparatus, it is desirable that a brief description of the method of working with the instrument should be given here.

Inside the mahogany case of the Electric Lamp there is a small brass vertical cylinder.

Take care that the piston works freely inside the cylinder, and that the cylinder is full of glycerine.

The object of this arrangement is to prevent the Lamp acting with violent jerks.

Also that the long vertical rod with the milled head at the top moves freely in the tube, and that the iron rods attached to the armature run freely to and fro in the helices of covered copper wire.

When the ends of the carbon rods recede too far from each other, an unsteady light is the result. This can be remedied by moving the small horizontal lever in the front of the large brass plate towards the right. By acting on a strong steel spring, this brings the points of the carbons nearer together.

If the carbon rods do not burn away equally, the light will not be opposite the centre of the parabolic reflector, or lens used for projecting it to a distance. A milled head on the right hand of the base of the Lamp acts on rackwork, which will regulate the height of the lower carbon rod. The upper rod will accommodate itself to the lower rod.

Owing to inequalities in the carbon, the electric light sometimes appears at the side or the back of the carbons. To set this right, move the upper rod by the ebonite head. When a parabolic reflector is used, the light should be brightest on the side towards the upright rod, but it should be brightest on the opposite side to the rod when a lens is used, or when the Lamp is placed in a lantern.

Before making contact with the battery by means of the turning lever of bright copper at the back of the Lamp, press the iron armature firmly forward, so that the iron rods are completely in the hollow coils or helices of covered copper wire.

INSTRUCTIONS FOR CHARGING THE BATTERY.

Fill the porous cells with nitric acid—that is, commercial aquafortis—and insert the platinum foil or carbon plate. In a strong stoneware vessel, mix one part of oil of vitriol—that is, commercial sulphuric acid—with seven parts of water. Fill the outer cells with this mixture, having first introduced the zinc

plates and porous cells. After the porous cells have been placed in the centre of the zinc plates, connect the platinum or carbon plate in each cell with the zinc plate in the next cell by means of the brass clamps; attach one of the clamps with the finger-screw at top to the unconnected platinum or carbon plate at one end of the battery, and the other clamp of the same kind to the unconnected zinc plate at the opposite end of the battery; then connect these ends with the copper wires as before directed. The battery will not attain its full power in less than half an hour after charging. When the battery is done with, the porous cells, zinc, and platinum or carbon plates should be well washed in water. The porous cells should be allowed to remain in fresh water for several hours.

Occasionally, when the zinc plates are taken out of the acid, a little mercury should be well rubbed over them, by means of a piece of rag tied round a small stick. This should be done before they are washed in water.

SETS OF APPARATUS FOR PRODUCING THE ELECTRIC LIGHT.

No. 1.—Small-size Electric Lamp with Reflector; 20 quart Bunsen's Cells; two Varnished Oak Trays, to hold 10 cells each; Carbon Rods.

No. 2.—Medium-size Automatic Electric Lamp; 30 quart Bunsen's Cells; three Varnished Oak Trays; Carbon Rods.

No. 3.—Large Automatic Electric Lamp; 50 quart Bunsen's Cells; five Varnished Oak Trays; Carbon Rods.

SPECTRUM APPARATUS, FOR SCREEN EXPERIMENTS.

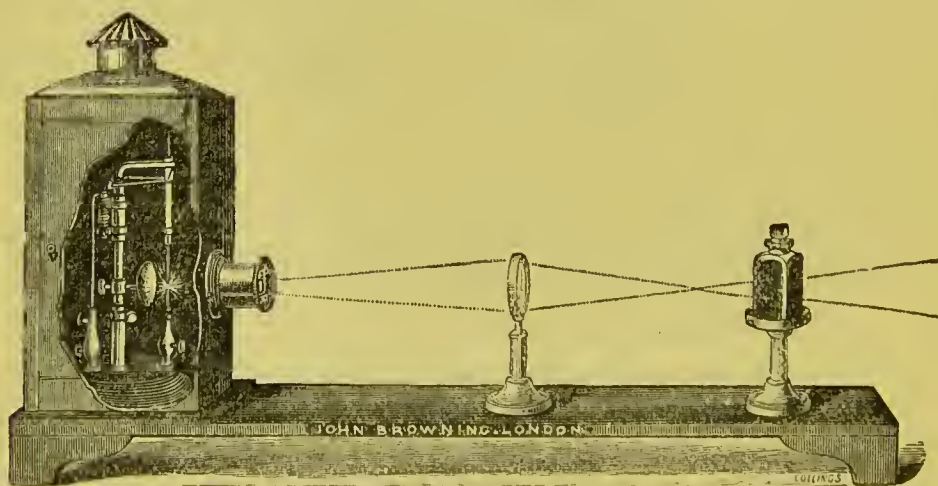


Fig. 20.

The engraving (Fig. 20) represents a new and complete set of apparatus, at a very low price, for projecting the spectra of metals, or the absorption bands of liquids, on a screen. The apparatus comprises an Automatic Electric Lamp and Lantern, with slit, a 20-quart Bunsen's cell, battery, and trays, mounted focussing lens, bisulphide of carbon prism and stand, platform for the whole, and packing case.

An inner case, which fits into the body of the Lantern, contains the Electric Lamp (Fig. 18) in packings, a set of chemicals which give the most brilliant spectra, and a supply of carbon rods and carbon crucibles.

A nozzle with lenses and $3\frac{1}{2}$ -inch condensers can be adapted to this Lantern, for showing diagrams or views on screen.

SCREEN SPECTRUM APPARATUS.

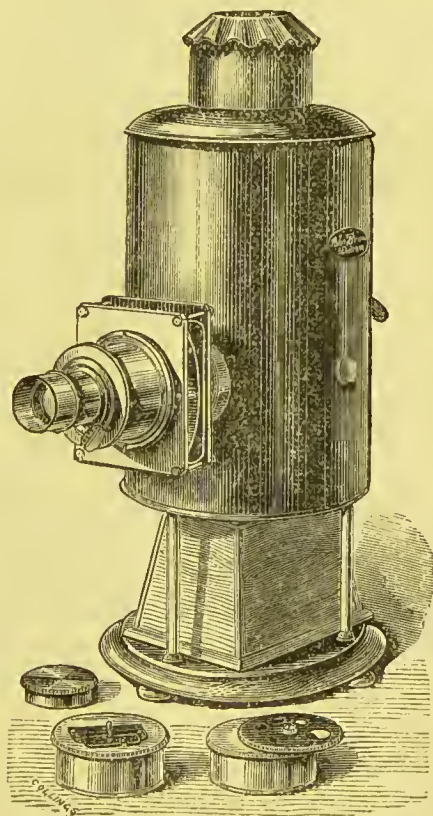


Fig. 21.

Fig. 21 represents a new metal body Electric Lantern for Spectroscopists, with an Electric Regulator specially adapted to the same. There are two nozzles, one for showing diagrams,

and the other for exhibiting spectra on a screen ; this apparatus is efficient in action, and yet economical in price.

The medium-size Automatic Electric Lamp, which is adapted to this Lantern, is of the best construction, and works well with from 20 to 30 pint Grove's cells, or the same number of quart Bunsen's.

SCREEN SPECTRUM APPARATUS.

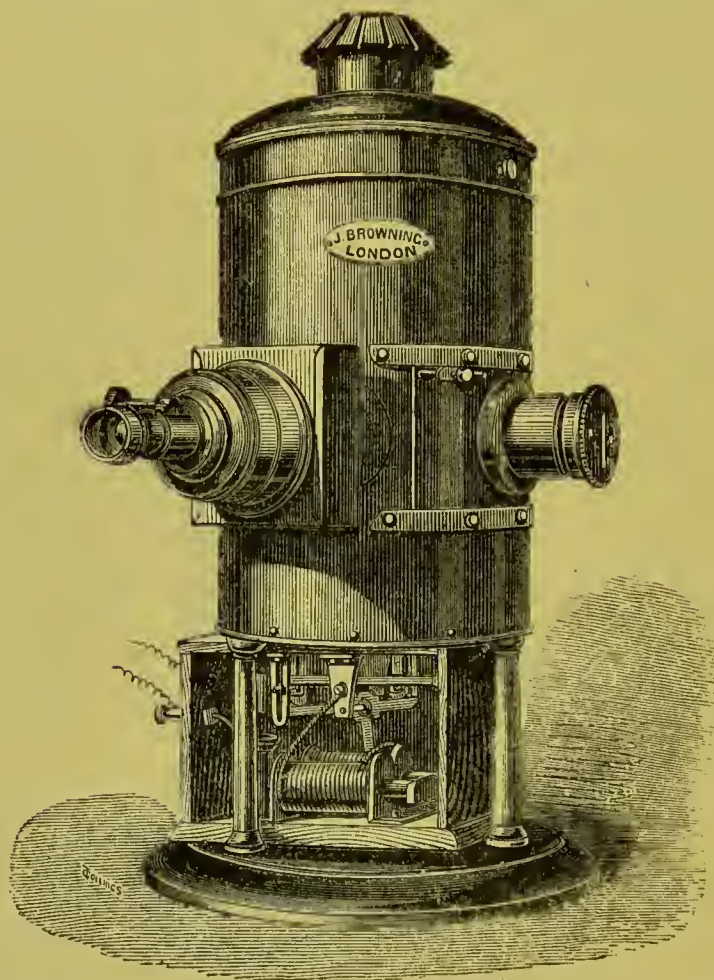


Fig. 22.

Improved Lantern, the body of brass, bronzed, with two nozzles, especially arranged for exhibiting spectra or diagrams on the same screen without shifting the Lantern or re-arranging the apparatus, with $3\frac{1}{2}$ -inch condensers (Fig. 22) ; a larger size Lantern is made with $4\frac{1}{2}$ -inch condensers.

A very complete set of Screen Spectrum Apparatus should consist of an Improved Electric Lantern, with $4\frac{1}{2}$ -inch con-

condensers (Fig. 22), the body of brass, bronzed, *with two fixed nozzles* specially arranged for exhibiting spectra or diagrams on the same screen, *without shifting the Lantern or re-arranging any part of the apparatus*; large-size Electric Regulator for the above, to work with from 25 to 50 Grove's or Bunsen's cells; two extra-size Bisulphide of Carbon Prisms; Prism Stand and Cover, adjustable for height with clamp motions; large Condensing Lens on Brass Stand, adjustable for height; Revolving Diaphragms Rotating Carbon Holder; Mahogany Case containing set of Metal; and Salts for burning in the Electric Arc, with carbon crucibles, pliers, and 6 feet of carbon rods, &c., for the large Regulator; Battery of 40 quart Bunsen's cells, in four varnished oak trays; set of Metals and Salts for burning in the Electric Arc, and showing their spectra; carbon crucibles, pliers, and carbon rods, in case.

HOW TO SHOW SPECTRA ON A SCREEN.

The apparatus for screen experiments consists of an Electric Lamp, a Lantern with a slit fixed vertically in place of condensers, a large convex lens mounted on a stand, and either one or two bottle prisms filled with bisulphide of carbon. This apparatus is arranged in the order shown in Fig. 20. The substance of which it is desired to obtain the spectrum should be placed in a hollow cup, bored out of the lower carbon rod in the Electric Lamp or Regulator. The upper rod should be brought into contact with the substance, and then carefully withdrawn until a steady spectrum, consisting principally of bright lines, is seen; the wider the two carbon rods are separated, so long as sufficient light is left, the better will be the result, as in this manner the brilliant continuous spectrum of the carbon rods, which masks the character of the bright lines, is got rid of. It must be clearly understood that although for all ordinary purposes it is essential that the Electric Lamp should be automatic, yet, for obtaining the spectra of any substance, *the Lamp must be fixed for a certain length of time, for so long as the poles are allowed to approximate to each other, as they would do by the automatic arrangement, so long we shall obtain only the continuous spectrum of the poles, as has been before described.*

The Lamp being arranged within the Lantern, as shown in Fig. 22, the slit on the nozzle of the Lamp must be placed in a vertical direction; the fixed lens, adjusted so that the height of the centre of the lens corresponds with the centre of the slit, must then be drawn backwards and forwards along the stand on which

the apparatus is fitted, until a sharp image of the slit is produced on a piece of paper, at a distance of from 10 to 20 feet, according to the size of the room in which the experiments are to be conducted ; the distance, however, should be no greater than will produce a spectrum sufficiently large to be seen by the audience. Having obtained the bright image of the slit in the manner just described, either one or two bottle prisms must be brought in front of the lens, just beyond the place at which the rays from the fixed lens cross, after coming nearly to a point. The prism must then be turned round slowly ; as it is turning, a spectrum will be seen to travel along the face of the screen, placed at an angle of nearly 90° to the apparatus, and at about the same distance from the apparatus at which the bright image of the slit was formed. With the small apparatus only one bisulphide of carbon prism can be used ; but with the large Electric Lamp (Fig. 19) and the Lantern (Fig. 22), the light is sufficiently strong to allow of the use of two prisms, and these will give a much longer spectrum. The small carbon crucibles made for the purpose are more efficient than an ordinary carbon rod for obtaining the spectra, as they will hold more of the material to be operated on.

Where time is an object, as in a lecture, a number of these crucibles, fitted in a series of tubes on a revolving holder on the centre round which these revolve, must be placed eccentrically, and the crucibles already charged with the various substances can be brought successively to coincide with the upper pole, and operated on at pleasure. It is difficult to obtain the spectrum of iron in this manner ; a better result can be secured by fixing in the upper carbon holder a piece of iron, and allowing this to come into contact with the carbon rod, taking care that they do not fuse together.

The most interesting experiment which can be made with this apparatus is the reversal of the soda line ; but the experiment is one which requires a little skill in manipulation, yet it is not difficult if the following precautions are attended to : Let the Lamp be carefully closed in with a cover of stiff brown paper over the apertures left for the purpose of ventilation ; set light to a piece of metallic sodium the size of a pea in a small iron spoon, by means of a spirit-lamp ; this must be done inside the Lantern, the spoon being first made nearly red-hot ; the spoon should be attached to a small holder, and left inside the Lamp, with the door shut, until the Lamp becomes filled with the vapour of the burning sodium. One of the carbon crucibles or the lower rod of the Electric Lamp having had a piece of

sodium placed on it, the upper end of the Electric Lamp can be brought into contact with it; the intense light generated will then produce a faint continuous spectrum on the screen, with a very bright sodium line predominant; after a short time, varying between a few seconds and two or three minutes, if the Lantern has been sufficiently filled with the vapour of the sodium burning in the spoon, these bright lines will become dark ones. Probably in the course of the experiments they will again become bright, and the reversal be repeated two or three times. The explanation of this experiment is, that the vapour of the sodium in the Lantern being cooler than the sodium burning between the poles of the Lamp, the bright light of the sodium is absorbed by the sodium vapour.

Salts of silver, zinc, copper, or small pieces of the metals themselves, give the most brilliant spectra.

This apparatus may be used without any modification for the purpose of showing the absorption bands in many coloured substances or liquids. In operating on liquids, they should be placed in a hollow wedge-shape cell; these should be brought in front of the slit, and moved along until such a thickness of the liquid comes in front of the slit as will produce sufficiently strongly-marked absorption bands in the spectrum. During these experiments the Electric Lamp must be kept burning continuously by the automatic action as steadily as possible. The liquids named in the series of specimens on page 49 all give good results. As a few of the most strongly-marked spectra, I will mention blood, sumach, permanganate of potash, chlorophyll, and magenta.

SPECTRUM APPARATUS FOR THE MICROSCOPE.

The writer has worked in conjunction with H. C. Sorby, Esq., F.R.S., in his experiments, having for their object the improvement of this apparatus, and has just perfected a new Micrometric arrangement, which possesses great advantages. The Micro-Spectroscope is applied to the eye-piece of a Microscope instead of an ordinary eye-piece.

It is applicable to opaque objects as well as transparent without preparation, and by its means two spectra may be compared at the same time with one lamp. It possesses the immense advantage over all other contrivances of the kind, that

the spectrum of the smallest object, or a particular portion of any object, may be obtained with the greatest certainty and

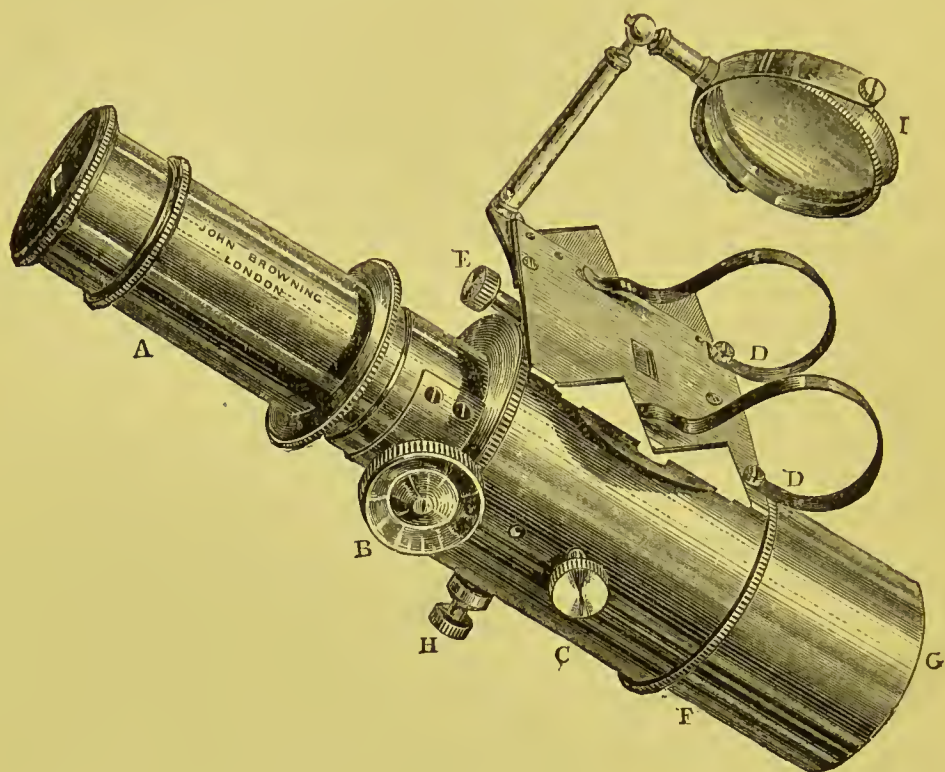


Fig. 23.

facility. This Micro-Spectroscope will indicate plainly the minutest quantity of blood, adulterations in wine, mustard, peppermint, oil, and many other articles of food, as well as the absorption bands in the leaves and juices of plants.

HOW TO USE THE MICRO-SPECTROSCOPE.

As will be seen from Fig. 23, the Micro-Spectroscope is a very compact piece of apparatus, consisting of several parts. The prism is contained in a small tube, A, which can be removed at pleasure. Below the prism is an achromatic eye-piece, having an adjustable slit between the two lenses; the upper lens being furnished with a screw motion to focus the slit. A side slit, capable of adjustment, admits, when required, a second beam of light from any object whose spectrum it is desired to compare with that of the object placed on the stage of the Microscope.

This second beam of light strikes against a very small prism suitably placed inside the apparatus, and is reflected up through the compound prism, forming a spectrum in the same field with that obtained from the object on the stage (Fig. 24, page 52). A (Fig. 23) is a brass tube carrying the compound direct-vision prism, and has a sliding arrangement for roughly focussing. B, a milled head, with screw motion to finally adjust the focus of the achromatic eye-lens.

C, milled head, with screw motion to open or shut the slit *vertically*. Another screw, H, at right angles to C, regulates the slit horizontally. This screw has a larger head, and when once recognized cannot be mistaken for the other.

D D, an apparatus for holding small tube, that the spectrum given by its contents may be compared with that from any other object on the stage.

E, a screw opening and shutting a slit to admit the quantity of light required to form the second spectrum. Light entering the aperture near E strikes against the right-angled prism which we have mentioned as being placed inside the apparatus, and is reflected up through the slit belonging to the compound prism. If any incandescent object is placed in a suitable position with reference to the aperture, its spectrum will be obtained, and will be seen on looking through it.

F shows the position of the field lens of the eye-piece.

G is a tube made to fit the Microscope to which the instrument is applied. To use this instrument, insert G, like an eye-piece in the Microscope tube. Screw on to the Microscope the object-glass required, and place the object whose spectrum is to be viewed on the stage. Illuminate with stage mirror if transparent, with mirror and lieberkühn and dark well if opaque, or by side reflector, bull's-eye, &c. Remove A, and open the slit by means of the milled head, H, at right angles to D D. When the slit is sufficiently open the rest of the apparatus acts like an ordinary eye-piece, and any object can be focussed in the usual way. Having focussed the object, replace A, and gradually close the slit till a good spectrum is obtained. The spectrum will be much improved by throwing the object a little out of focus.

Every part of the spectrum differs slightly from adjacent parts in refrangibility, and delicate bands or lines can only be brought out by accurately focussing their own parts of the spectrum. This can be done by the milled head, B. Disappointment will occur in any attempt at delicate investigation if this direction is not *carefully attended to*.

When the spectra of very small objects are to be viewed, powers of from $\frac{1}{3}$ -inch to $\frac{1}{8}$ th may be employed.

Blood, madder, aniline dyes, permanganate of potash solution, are convenient substances to begin experiments with. Solutions that are too strong are apt to give dark clouds instead of delicate absorption bands.

Small cells or tubes should be used to hold fluids for examination.

Objects, such as crystals, should invariably have a small cardboard diaphragm, $\frac{1}{8}$ diameter, placed beneath them; the spectrum is then much better defined. With a slide containing a mass of small crystals, the object need merely be thrown a little out of focus. When observing the spectra of liquids in experiment cells, or through small test-tubes, always slip over the tube containing the $1\frac{1}{2}$ or 2-inch objective a cap with a hole $1\text{-}16$ th of an inch diameter. Slide the tube just sufficiently to bring the small hole a little within the focus of the objective. By this arrangement all extraneous light is prevented from passing up the body of the Microscope, except what passes through the object. Unless this precaution be attended to, a false result is sometimes obtained.

Substances which give bands or lines in the red, are best seen by gaslight, while those which give bands in the blue are brought out far better by daylight. Such a specimen as oxalate of chromium and soda is almost opaque by daylight, showing no bands, though, when examined by a lamp, the spectrum exhibits three beautifully fine lines in the red, two of which are exceedingly delicate. Again, uranic acetate can only be seen to advantage by strong daylight, since the band in the violet would be invisible by lamp-light.

A number of dyes are beautifully shown by being dissolved in gelatine. A plate containing one dozen small strips of gelatine about $\frac{1}{4}$ -inch wide and $\frac{3}{4}$ -inch long, is exceedingly convenient for the purpose of showing the spectra of these dyes. When the slit of the Spectroscope is placed across the junction of two of the plates, any two spectra can be seen at the same time in the field of view, and thus comparisons may be made between them. If two such plates be superimposed, a great number of spectra, in which the absorption bands of two substances appear at the same time, are shown. These plates are much more easy to manipulate with than tubes.

For information as to mapping spectra with the Micro-Spectroscope, see the instructions under a separate heading.

OBJECTS FOR THE MICRO-SPECTROSCOPE.

Liquids in Glass Tubes Hermetically Sealed, as used by
H. C. SORBY, Esq., F.R.S.

CLASS I.

Specimens for Illustrating the Application of the Micro-Spectroscope to Chemistry.

PERMANENT SPECIMENS.

- | | |
|-----------------------|----------------------|
| 1. Cruentine Neutral. | 7. Chromic Oxal. |
| 2. Didymium. | 8. Cruentine Acid. |
| 3. Aniline Product. | 9. Hypericine. |
| 4. Hæmatin Deox. | 10. Uranium Sulph. |
| 5. Uranium Carb. | 11. Cobalt Chloride. |
| 6. Purpurine. | 12. Cobalt Cyanide. |

In Morocco Case.

CLASS II.

STANDARD BLOWPIPE BEADS.

For Spectrum Tests of Metallic Oxides.

- | | |
|--------------------|------------------------|
| 1. Cobalt (blue). | 6. Tungsten. |
| 2. Erbium. | 7. Didymium. |
| 3. Uranium. | 8. Chromium. |
| 4. Molybdenum. | 9. Copper (red). |
| 5. Uranium Borate. | 10. Uranite (Mineral). |

The Spectra are best seen with Lamplight.

CLASS III.

Dyes.

Consisting of a plate containing a set of *twelve* substances in gelatine in a most compact form, and so arranged that *two* spectra can be seen simultaneously by placing the plate in front of the slit of the Spectroscope, giving in all eighteen different comparisons. Also, by superposing two such plates at right angles, thirty-six compartments are visible which show the phenomena of mixed spectra.

This arrangement is exceedingly useful for showing absorption spectra at a lecture, since six spectra can be thrown on to the screen at once.

The Plates are fitted in Morocco Cases.

HOW TO MAKE A MAP OF A SPECTRUM WITH A STUDENT'S SPECTROSCOPE.

Place the eye-piece with cross-wires in the telescope, with the cross in the direction of an λ . Then move the telescope so that the point where the wires bisect comes successively in contact with the various lines, noting the readings of the vernier on the arc. From these readings, by the help of any mechanical scale of equal parts, a map may be easily constructed.

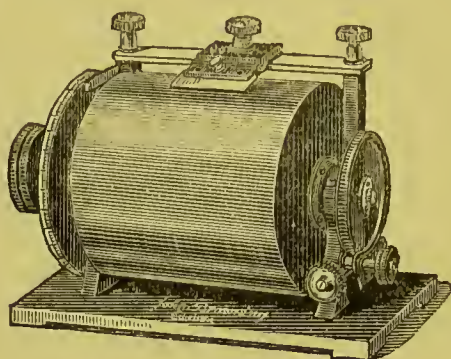


Fig. 26.

HOW TO MAP A SPECTRUM WITH BECKLEY'S SPECTROGRAPH.

Place a sheet of paper on the metal cylinder of the Spectrograph. Note the position of any line in a spectrum, and set the divided edge of the Spectrograph to the corresponding division on the vernier. Draw a line on the paper along the steel straight-edge. Now take the reading of another line, and proceed in a similar manner until the map is completed.

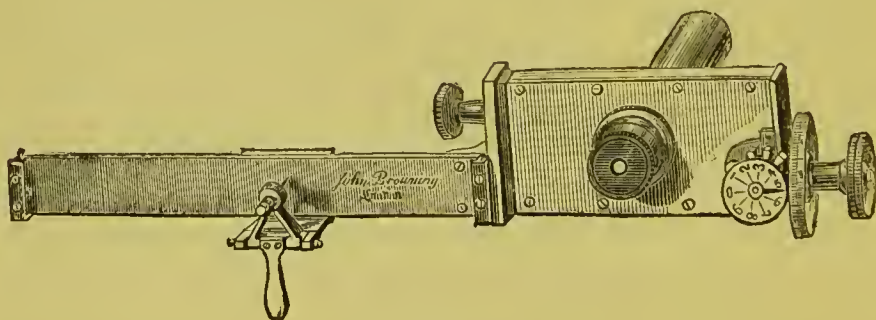


Fig. 27.

HOW TO MAP A SPECTRUM WITH COLONEL CAMPBELL'S SPECTROGRAPH.

Place a strip of smoked glass in the frame attached to the side of the Micrometer (Fig. 27). Place the tube of the Micrometer in the eye-draw of the telescope of a Spectroscope. Bring the cross-wires of the Micrometer on to one of the lines in the spectrum by turning the Micrometer screw. Now draw a line on the smoked glass with the small ruling machine on the left hand in the engraving. Repeat this operation for as many lines as it is desired to map, *always turning the screw in one direction*. The smoked glass may be varnished and used as a negative to print copies from by photography, on glass or paper, at pleasure.

HOW TO MAP A SPECTRUM WITH THE AUTOMATIC SPECTROSCOPE.

Place the Filar Micrometer in the telescope. Bring some easily recognized line to correspond with the fixed wire of the Micrometer; then, by moving the Micrometer-head, make the movable wire coincide with another line in the spectrum. Read the indication on the Micrometer-head, and note it. The small wheel at the side of the Micrometer-head shows whole revolutions of the screw, the long divisions on the Micrometer-head tenths, and the short divisions hundredths of a revolution. Then take a scale of equal parts, and represent each division on the Micrometer-head by one division on the scale.

It is obvious that the map of a spectrum may be made of any size by varying the scale from which the divisions are taken.

HOW TO MAP A SPECTRUM OF ABSORPTION BANDS WITH A MICRO-SPECTROSCOPE.

Fig. 24 represents the upper part of the Micro-Spectroscope. Attached to the side is a small tube, A A. At the outer part of this tube is a glass plate, blackened, with a fine clear white pointer in the centre of the tube. The lens, C, which is focussed by moving the small studs at M, produces an image of the bright pointer in the field of view by reflection from the surface of the prism nearest the eye. On turning the Micrometer screw, M, the slide which holds the glass plate is made to travel in grooves, and the fine pointer is made to traverse the whole length of the spectrum.

It might at first sight appear as if any ordinary spider's web

or parallel wire Micrometer might be used instead of this contrivance; but on closer attention it will be seen that as the spectrum will not permit of magnification by the use of lenses, the line of such an ordinary Micrometer could not be brought to focus and rendered visible. The bright pointer of the new arrangement possesses this great advantage—that it does not illuminate the whole field of view.

If a dark wire were used and illuminated, the bright diffused light would almost obscure the faint light of the spectra, and entirely prevent the possibility of seeing, let alone measuring, the position of lines or bands in the most refrangible part of the spectrum.

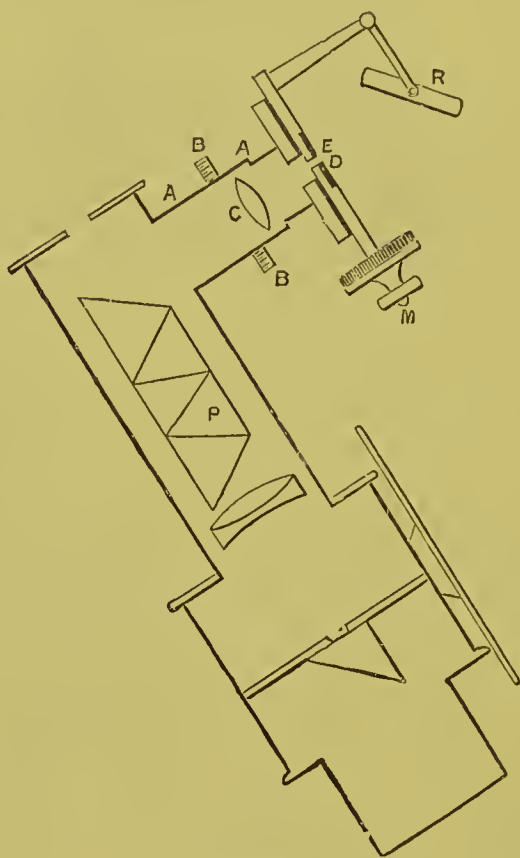


Fig. 24.

To produce good effects with this apparatus, the upper surface of the compound prism, P, must make an angle of exactly 45° with the sides of the tube. Under these circumstances, the limits of correction for the path of the rays in their passage through the dispersing prisms are very restricted, and must be strictly observed. The usual method of correcting by

the outer surface is inadmissible. For the sake of simplicity, some of the work of the lower part of the Micro-Spectroscope is omitted in the engraving. As to the method of using this contrivance: with the apparatus just described, measure the position of the principal Fraunhofer lines in the solar spectrum. Let this be done *carefully*, in *bright* daylight. A little time given to this measurement will not be thrown away, as it will not require to be done again. Note down the numbers corresponding to the position of the lines, and draw a spectrum from a scale of equal parts. About three inches will be found long enough for this spectrum; but it may be made as much longer as is thought desirable, as the measurements will not depend in any way on the distance of these lines apart, but only on the micrometric numbers attached to them. Let this scale be done on cardboard, and preserved for reference. Now measure the position of the dark bands in any absorption spectra, taking care for this purpose to use lamplight, as daylight will give, of course, the Fraunhofer lines, which will tend to confuse your spectrum. If the few lines occurring in most absorption spectra be now drawn to the same scale as the solar spectrum, on placing the scales side by side, a glance will show the exact position of the bands in the spectrum relatively to the Fraunhofer lines, which thus treated form a natural and unchangeable scale (see diagram, Fig. 24*). But for purposes of comparison it will be found sufficient to compare the two lists of numbers representing the micrometric measures, simply exchanging copies of the scale of Fraunhofer lines, or the numbers representing them will enable observers at a distance from each other to compare their results, or even to work simultaneously on the same subject.

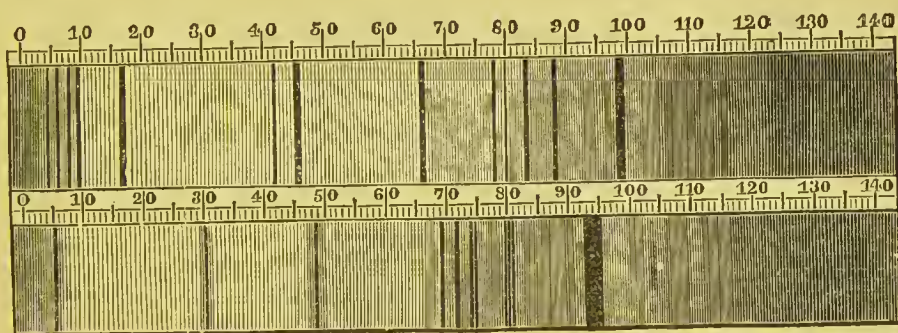


Fig. 24*.

It is a great advantage of this contrivance that it does not monopolize one of the two spectra, as is the case with the use of the quartz scale; for in describing two spectra only slightly

differing from each other, it may be used at once to determine the difference between them. Many substances give two different spectra when examined by transmitted or reflected light, though there is generally a close resemblance between them.

Dr. MacMunn, in his work, *The Spectroscope in Medicine*, has thus described a Photographed Scale Micrometer which I first applied to the Microscope at his suggestion :—

"Mapping by means of Photographed Scale.—Mr. Browning has fixed to the side of the prism tube of my Micro-Spectroscope a tube in the position of A (Fig. 11), which contains a photographed scale illuminated by a small mirror, and which is capable of being focussed by a small lens in the position of C, so that, on looking into the instrument, one can see the spectrum beautifully and accurately divided into 100 equal parts, since the image of the scale is reflected to the eye from the surface of the uppermost prism. By means of this scale readings can be made at once; the only precaution necessary is to see that D, or the sodium line if D cannot be got, always stands at the same number on the scale. To map absorption spectra measured on this scale is exceedingly simple; all we have to do is to lay down a line as many millimetres long as there are divisions in the scale, and mark the position of the bands on this line. Of course, as in the case of the bright point Micrometer, a map of the solar spectrum has to be first made. Another great advantage of this scale, besides the rapidity with which measurements can be made with it, is this, that it enables us to record certain data from which maps can at any future time be constructed. Thus we can record a spectrum somewhat in this manner :—

Extent	10 to 70	
First band	...	15 to 18	... (A)
Second band	...	25 to 30	... (B)

Any one who has worked with the Spectroscope can appreciate the convenience of this method."

WORKS ON SPECTRUM ANALYSIS.

THE SPECTROSCOPE AND ITS WORK. By RICHARD A. PROCTOR, Author of "Saturn and its System," "The Sun," "The Moon," &c. With Coloured Diagram of the Spectra, and many Illustrations. Fcap. 8vo., limp cloth, 1s. (A popular, though by no means superficial, account of the Spectroscope and its marvellous work. For general readers, and as a text book.)

SPECTRUM ANALYSIS in its Application to Terrestrial Substances and the Physical Constitution of the Heavenly Bodies, familiarly explained by Dr. H. SCHELLEN, Director der Realschule I. O. Cologne. Translated from Second German Edition by JANE and CAROLINE LASSELL. Edited, with Notes, by WILLIAM HUGGINS, LL.D., D.C.L., F.R.S. In 1 vol., 8vo., with 13 Plates (6 Coloured), including Angström's and Kirchoff's Maps and 223 Woodcuts. Price 28s. cloth.

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A Coloured Chart of the Spectra of the Metals	0	7	0
A Coloured Chart of the Spectra of the Stars and Nebulæ... ..	0	7	0
Any of the above, mounted on cloth and roller, and varnished	0	10	6

NOTE.—The Numbers in this List of Prices correspond with the Numbers appended to the Engravings of the Instruments in the body of the Book.

JANUARY, 1883.

The following Prices are Nett for Cash ; Half-price allowed for returned Packages, if Carriage Paid.

ORDERS SHOULD BE ACCOMPANIED BY A REMITTANCE.

DIRECT-VISION SPECTROSCOPES.

THE MINIATURE SPECTROSCOPE.

This Instrument will show many of the Fraunhofer lines, the bright lines of the metals and gases, and the absorption bands in coloured gases, crystals, or liquids.

Miniature Spectroscope, with plain slit	£1 2 0
Miniature Spectroscope, with adjustable slit (Fig. 1)	1 13 0
Miniature Spectroscope, in Brass Case, with glass at each end, for use in the Bessemer process	1 16 6
Morocco Case, extra	0 2 0
Miniature Spectroscope and adjustable slit, with Achromatic Lenses, in Morocco Case	2 6 0
Rain-band Spectroscope as recommended by Mr. Rand Capron, in Morocco Case (Fig. 2)	2 10 0
Rain-band Spectroscope, with fixed slit, without glass caps or case	1 10 0
Browning's new Miniature Spectroscope, with Micrometer, price, complete in Case (Fig. 3)	3 10 0

This portable and complete Instrument may be used for showing any of the leading experiments in Spectrum Analysis ; the Fraunhofer Lines ; the Lines in the Spectra of the Metals, and the Alkaline Earths and Alkalies ; the Spectra of Gases ; and Absorption Bands.

Applied to a Telescope, it may be used for viewing the Lines of the Solar Prominences. It can also be used as a Micro-Spectroscope. The position of the Bands in any Spectrum may be measured with the attached Micrometer.

Clip Stand for Miniature Spectroscope	£0 9 6
New Pocket Direct-Vision Spectroscope, in Morocco Case (Fig. 4)	4 10 0
New Form Direct-Vision Spectroscope, with five prisms, fitted in Mahogany Case	5 15 0

This Spectroscope is a most powerful and portable Direct-Vision Instrument, easily separating the D lines in the Solar Spectrum.

Extra power for the above	£0 12 6
Large size, higher dispersive power, and extra power eye-piece, complete in Mahogany Case...	6 18 0

Portable Tripod Clip Stand for the above	£1 15 0
Direct-Vision Solar Spectroscope, for use with a Telescope for viewing the bright lines or the forms of the Solar Prominences	8 10 0

CHEMICAL SPECTROSCOPES.

The Student's Spectroscope, in Stained Cabinet (Fig. 5)	£6 10 0
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This Instrument has a prism of extremely dense glass of superior workmanship. The circle is divided, and reads with a vernier, thus dispensing with the inconvenience of an illuminated scale; this arrangement possesses the very great advantage of giving angular measures in place of a perfectly arbitrary scale.

The slit is also furnished with a reflecting prism, by means of which two spectra can be shown in the field of view at the same time.

The Instrument is so arranged that, with a slight alteration of the adjustments, it can be used for taking the refractive and dispersive powers of solids or liquids.

The Model Spectroscope (Fig. 6), with two prisms, in Polished Mahogany Cabinet	£15 0 0
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This Instrument has two dense glass prisms, two eye-pieces, rack motion to Telescope, and tangent screw motion to vernier. It will widely separate the D lines.

Photographed Micrometer to either the one or two prism Spectroscope	£1 15 0
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The Model Spectroscope, with four prisms, in superior Cabinet, with fittings and two eye-pieces	27 10 0
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This Instrument is guaranteed to show the Nickel line between the D lines in the solar spectrum.

Browning's Automatic Action, extra	£6 10 0
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LARGE TABLE SPECTROSCOPES.

The Large Model Spectroscope, for the use of Physicists, made on the plan of the Gassiot Spectroscope, in Polished Mahogany Cabinet	£38 10 0
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This Instrument has four very large dense glass prisms and Telescope with object-glasses $1\frac{1}{2}$ in. diameter, and 18 in. focal length, furnished with three eye-pieces. It will show two or three lines between the D lines in the solar spectrum. Any smaller number of the prisms can be used when desired.

Dividing ditto on Silver, extra	£2 0 0
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Browning's Automatic Action to the above	12 10 0
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The above Instrument, the circle divided on Silver to 10 seconds, with five prisms, four eye-pieces, and parallel wire Micrometer, for measuring the position of lines to $\frac{1}{10000}$ of an inch; the whole in Mahogany Case	55 10 0
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Browning's Automatic Action to the above	15 0 0
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BROWNING'S UNIVERSAL AUTOMATIC SPECTROSCOPE.

In this Instrument the prisms are automatically adjusted to the minimum angle of deviation for the particular ray under examination; the position of the lines in the spectrum is measured by means of a Micrometer; the revolution of this Micrometer screw adjusts the prisms automatically for the ray under measurement. The Instrument has six prisms, and by means of the reversion of the ray a dispersive power of 12 prisms is obtained. By changing the position of one of the prisms, any dispersive power from 2 to 12 prisms can be used at pleasure, without deranging any of the adjustments of the Instrument. The Instrument is therefore applicable to every class of spectrum work either in the Laboratory or Observatory.

Price of the Universal Automatic Spectroscope, with six prisms, best Filar Micrometer and Battery of 9 Eye-pieces, in Mahogany Cabinet complete (Fig. 7)	£65 0 0
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The same Instrument with light framework adapted for application to an Astronomical Telescope	£70	0	0
Adapter, with movement of rotation, for attaching the Automatic Spectroscope to a Telescope	6	10	0
Adapter, with two slides, and mechanical motions, to enable the observer to set the Spectroscope at any degree of eccentricity to the Solar disc, so as to sweep either round the Sun's limb to search for prominences, or in the neighbourhood of the Chromosphere	15	0	0
Browning's Universal Automatic Spectroscope, 6 prisms, with the ray inverted, giving a dispersive power of 12 prisms, with the prisms and object-glasses of the Telescopes $1\frac{1}{2}$ in. diameter, and Telescopes 18 in. focal length, Filar Micrometer, and 9 eye-pieces, &c., &c. ...	150	0	0

BROWNING'S AUTOMATIC SOLAR SPECTROSCOPE.

Dr. Henry Draper's important discovery of the presence of oxygen in the Sun, described in *Nature*, No. 409, August 30, 1877, will direct renewed attention to the Solar Spectrum.

The Automatic Solar Spectroscope (Fig. 9) will show the Solar Spectrum with exquisite definition, and if attached to the eye-piece of a Telescope of 3 inches or more in diameter, it will show the form of the Solar Prominences.

As this Spectroscope can be used with any dispersive power from 2 to 10 prisms, it can be arranged for observing the Spectra of the Stars and Nebulæ. Without a Telescope it can be employed for any kind of work in Spectrum Analysis.

By means of the reversion of the ray this Spectroscope gives a dispersive power equal to 10 prisms, and this dispersive power may be changed at pleasure by the observer. This Instrument is very light, and can be adapted to a Telescope as small as 3 inches in aperture. It is provided with a movement of rotation for searching for Solar Prominences.

Complete in Case, with Eye-pieces	£28	0	0
Rack Adjustment to the rotary motion, extra	2	10	0

By means of the reversion of the ray this Spectroscope gives a dispersive power equal to 10 prisms, and this dispersive power may be changed at pleasure by the observer. It is well adapted for use with any Telescope, either a Reflector or Refractor, from 6 inches to 12 inches in aperture (Fig 10).

Price complete, with Set of four Eye-pieces	£42	10	0
Table Stand for using the Automatic Spectroscope above described, without an Astronomical Telescope for viewing the Spectra of Metals, Salts, or Gases	1	15	0

STAR SPECTROSCOPES.

MCCLEAN'S NEW STAR SPECTROSCOPE (PATENT).

The Star Spectroscopes in general use are expensive and difficult to manipulate with. This arises from the fact that, in most instruments, the image of a star is required to fall within the jaws of a narrow slit, not more than $\frac{3}{1600}$ inches in width, and an equatorially mounted telescope with clockwork is almost indispensable for using them. Star Spectroscopes of simpler construction, both with and without cylindrical lenses, have been made, but their performance has not been found satisfactory. In the Instrument contrived by Mr. McClean (Fig. 11), exquisitely fine lines can be seen in the spectra of stars without the use of any slit.

Price of the Spectroscope, in Morocco Case	£2 10 0
Adjustable Slit and Convex Lens, to be used as an addition to McClean's Star Spectroscope, for showing the Fraunhofer lines in the Solar Spectrum, the bright lines of the metals, alkalies, gases, &c., <i>extra</i> ...	0 18 6
McClean's Spectroscope, for showing both astronomical and chemical spectra, in Case complete... ..	3 7 6
Rotating Telescope Nozzle, as used by Mr. McClean to carry the New Star Spectroscope, and an astronomical eye-piece, or two eye-pieces of different powers	1 15 0

This contrivance greatly facilitates the use of McClean's Star Spectroscope, the Spectroscope being carried by one arm, and an eye-piece for observing by the other. Any star seen in the eye-piece can in an instant be examined with the Spectroscope; or a low power may be used in one arm for finding an object, and a high power for observing it, to be changed rapidly without unscrewing.

STAR SPECTROSCOPES.

Star Spectroscope, with 1 prism, packed in Polished Mahogany Case ...	£8 8 0
Star Spectroscope, with 2 prisms, reflecting prism, to show two spectra at once, and Micrometer Measuring Apparatus for Mapping Spectra, packed in polished Mahogany Case (Fig. 12)	14 0 0
Insulated Spark Apparatus attached to mirror, for obtaining the spectra of the metals for comparison, adapted to either of the above Instru- ments	1 1 0
Star Spectroscope of the best construction, with adjustable reflecting prism and mirror, with finest object-glass, Micrometric Apparatus for Measur- ing the Lines of the Spectrum to $\frac{1}{10000}$ of an inch, extra eye-piece, and ivory tube to reader of vernier, as made for Dr. W. Huggins, F.R.S., packed in polished Mahogany Case, with Insulated Spark Apparatus complete (Fig. 12)	21 0 0
The Amateur's Star Spectroscope, in Mahogany Case (Fig. 13)	4 0 0
Browning's Bright Line Micrometer, for measuring the position of bright lines in the spectra, adapted to the above	2 5 0

INDUCTION COILS,

For working Induction Tubes or obtaining the Spectra of the Metals by the aid of the Electric Spark.

Mr. JOHN BROWNING begs to inform scientific gentlemen that the adoption of an improved method of winding Induction Coils has enabled him to increase their efficiency and reduce their cost. Every Coil is guaranteed to give the length of spark named.

FOR WORKING INDUCTION TUBES.

Induction Coil, to give a $\frac{1}{2}$ -in. spark in dry air, with 1 quart-size Bunsen's Cell (Fig. 14)	£3 5 0
Induction Coil, to give a 1-in. spark in dry air, with 1 quart-size Bunsen's Cell	5 10 0
Induction Coil, to give a $1\frac{1}{2}$ -in. spark in dry air, with 2 quart-size Bunsen's Cells	6 15 0

FOR SPECTRUM ANALYSIS.

Induction Coil, to give a $2\frac{1}{2}$ -in. spark in dry air, with 3 quart-size Bunsen's Cells	£10 0 0
Induction Coil, to give a $3\frac{1}{2}$ -in. spark in dry air, with 3 quart-size Bunsen's Cells	12 15 0
Induction Coil, to give a $4\frac{1}{2}$ -in. spark in dry air, with 5 quart-size Bunsen's Cells	16 0 0
Induction Coil, to give a 6-in. spark in dry air, with 6 quart-size Bunsen's Cells	22 0 0

Where the trouble of charging Bunsen's Cells is objected to, or it is desirable to avoid the nitrous fumes they give off, Bichromate Batteries can be supplied. These Batteries are very cleanly, but not nearly so powerful as the Grove's or Bunsen's Batteries, so that the Coils will not work with their full power when they are used.

Bichromate Battery, with 4 Cells, arranged so that, by using a winch, the elements may be removed from the exciting solution at pleasure. These Batteries may be used many times without re-charging.

Bichromate Battery, 4 large Cells, lifting elements	£4 10 0
Bichromate Battery, 6 large Cells, lifting elements	6 10 0

BROWNING'S SPARK CONDENSER.

This contrivance is designed to replace the Leyden Jars which are generally used with Induction Coils to increase the temperature of the spark when it is required for spectrum analysis. The apparatus consists of an arrangement of ebonite plates coated with tinfoil, and enclosed in a mahogany case. Any amount of surface may be used at pleasure by rotating the circle on the top of the case. Unlike the Leyden Jar, the action of the apparatus is not affected by damp. A very convenient arrangement for holding the metals of which the spectra are required, screws on to the lid of the case, and when not in use packs inside the lid (Fig. 15).

Becquerel's Apparatus for obtaining continuous spectra from solutions of salts of the metals, is attached when required.

Price of the Spark Condenser for Coils, giving a $2\frac{1}{2}$ -inch spark, with levers for changing the number of Plates	£2 15 0
Price of the Spark Condenser for Coils up to 5-inch spark, with levers for changing the number of Plates	3 15 0
Price of the Spark Condenser for Coils giving 6-inch to 8-inch spark	9 10 0
Becquerel's Apparatus extra	0 15 0

BESSEMER SPECTROSCOPES.

Either of the Direct-Vision Spectroscopes, enumerated on pages 12 and 13, are well adapted for viewing the Bessemer flame, and great numbers are in constant use for this purpose in all parts of Europe, the highest power being best adapted for the purpose; but Mr. Browning has devised a special Instrument of very great dispersive power, having an eye-piece of large field, which shows the whole of the spectrum, giving admirable definition in all parts of the field.

Direct-Vision Bessemer Spectroscope, with 10 prisms, complete in Mahogany Case ... | ... | ... | ... | £12 10 0 |

The Bessemer Spectroscope (Fig. 16) is a much more powerful Instrument. The Telescope has a motion between pivots, near the top of the case. There are cross-wires in the field of view to assist the observer in concentrating his own attention, or directing that of others, to any particular line in the spectrum. This Instrument

is so contrived that the back of the observer is turned to the brilliant flame, which renders vision much easier. A condensing lens, shown in the engraving, which works on a rod in front of the slit, can be fixed so as to produce an image of the flame on the slit; by moving the Instrument about on the hinged joint and swivel, the spectrum of any portion of the flame can be examined at pleasure. The slit is protected from the action of dust by means of a glass cover, and when the Instrument is not in use it can be unscrewed and enclosed in the box.

Price of the Instrument complete, with Stand £25 0 0

BROWNING'S ELECTRIC REGULATORS.

Small Electric Regulator, with parabolic reflector (Fig. 17). This Regulator will give a powerful and steady light, with from 10 to 20 quart-size Grove's or Bunsen's Cells. Price £2 5 0

Small Electric Regulator (Fig. 18), without reflector, for use in the Lantern, with adjustment for keeping the points of the burning carbons at one height, or separating them to any required distance. This adjustment is indispensable for projecting the spectra of burning metals on a screen. With 20 quart-size Bunsen's Cells, this Regulator will illuminate a 10-foot disc. Price 2 15 0

Parabolic Reflector, if required, extra 0 6 6

In these Regulators, described in p. 38, both carbons are moved by the electricity of the battery employed (without the aid of clockwork); the light remains uniform in height and more steady in action than any of the expensive Regulators previously introduced.

Medium-size Automatic Electric Regulator. This Lamp works well with from 20 to 30 pint Grove's Cells, or the same number of quart Bunsen's (as Fig. 19). Price £7 10 0

Parabolic Reflector, extra 1 15 0

Large-size Automatic Regulator. From 25 to 50 quart Grove's Cells, or the same number of 2-quart Bunsen's, should be used with this Lamp (Fig. 19). Price 9 9 0

Parabolic Reflector 2 2 0

Carbon Rods, for burning in small Lamp per foot 0 1 0

Ditto ditto per dozen feet 0 10 0

Large Rods, for burning in the large Lamp per foot 0 1 6

Ditto ditto per dozen feet 0 15 0

Carbon Cups, for holding metals to obtain their spectra, small, per dozen 0 10 6

Large ditto ditto ditto ditto 0 12 0

GALVANIC BATTERIES.

Grove's Cells pints (reputed) 11/, quarts (reputed) £0 14 0

Bunsen's Cells pints 5/6, quarts 6/6, 2 quarts 0 9 0

Varnished Oak Trays for 6 cells 6/, for 10 cells 0 10 6

Bichromate Battery, 4 lifting cells large size £4 10, 6 cells 6 10 0

Insulated Copper Wire... .. per yard, 2d. to 0 1 9

Porous Cells per dozen pints 10/, quarts 0 14 0

Stoneware Cells... .. „ pints 12/, quarts 16/, 2 quarts 1 0 0

Carbons „ pints 12/, quarts 16/, 2 quarts 0 18 0

Zincs „ pints 12/, quarts 18/, 2 quarts 1 4 0

Bichromate of Potash „ „ „

BROWNING'S SPECTRUM APPARATUS, FOR SCREEN EXPERIMENTS.

JOHN BROWNING has great pleasure in introducing to the notice of Lecturers and others a New and Complete Set of Apparatus, at a very low price, for projecting the spectra of metals, or the absorption bands of liquids, on a screen. The Apparatus comprises an Automatic Electric Lamp and Lantern, with slit, a Bunsen's battery of 20 quart-size cells, and trays, mounted focussing lens, bisulphide of carbon prism and stand, platform for the whole, and packing case £12 10 0

An inner case can be supplied, which fits into the body of the Lantern, contains the Electric Lamp (Fig. 18) in packings, a set of chemicals which give the most brilliant spectra, and a supply of carbon rods and carbon crucibles ... £4 0 0

Price of the Complete Set of Spectrum Apparatus, packed, £16 10s.

Nozzle with lenses and $3\frac{1}{2}$ -inch condensers, for showing diagrams or views

on screen, extra £2 15 0

Nozzle with $4\frac{1}{2}$ -inch condensers, extra 4 10 0

SCREEN SPECTRUM APPARATUS.

JOHN BROWNING begs to inform Spectroscopists that he has just introduced a new metal body Electric Lantern, with an Electric Regulator specially adapted to the same, for showing diagrams, and exhibiting spectra on a screen; thoroughly efficient in action, and yet economical in price.

The Automatic Regulator is of the best construction, exactly similar to his now well-known large Regulator, but arranged to burn with a smaller number of cells.

Price of the Electric Lantern, for medium-size Lamp, with metal body,

japanned bronze green, with two nozzles, interchangeable—one for showing diagrams, with $3\frac{1}{2}$ -inch condensers, the other for spectrum

analysis (Fig. 21) £7 10 0

MEDIUM-SIZE AUTOMATIC ELECTRIC LAMP

FOR THE ABOVE LANTERN.

This Lamp works well with from 20 to 30 pint Grove's cells, or the same

number of quart Bunsen's (Fig. 19) £7 10 0

LARGE ELECTRIC LANTERNS.

Improved Lantern, the body of brass, bronzed, *with two nozzles*, specially arranged for exhibiting spectra or diagrams on the same screen *without shifting the Lantern or re-arranging the apparatus*, with $3\frac{1}{2}$ -inch

condensers (Fig. 22) £11 10 0

Ditto, ditto, larger size, and $4\frac{1}{2}$ -inch condensers 16 10 0

Large Electric Lamp (Fig. 18) for either of the above Lanterns 9 9 0

VERY COMPLETE SET OF SCREEN SPECTRUM APPARATUS.

Improved Electric Lantern, with $4\frac{1}{2}$ -inch condensers, the body of brass, bronzed, with two nozzles specially arranged for exhibiting spectra or diagrams on the same, screen, without shifting the Lantern or re-arranging the apparatus. Large-size

Electric Regulator for the above, to work with from 25 to 50 Grove's or Bunsen's cells. Two extra-size Bisulphide of Carbon Prisms. Prism Stand and Cover, adjustable for height with clamp motions. Large Condensing Lens on Brass Stand, adjustable for height. Revolving Diaphragm. Rotating Carbon Holder. Mahogany Case containing set of Metals and Salts for burning in the Electric Arc, with carbon crucibles, pliers, and 6 feet of carbon rod, &c., for the large Regulator. Battery of 40 quart Bunsen's Cells in four varnished Oak Trays.

Price of the set of Apparatus, complete £50 0 0

Browning's New Lantern Microscope, adapted to the above, for showing

Microscopic Objects on a screen 5 10 0

Set of Metals and Salts for burning in the Electric Arc, and showing their spectra, with carbon crucibles, pliers, and 5 feet of carbon rod, in

Mahogany Case 2 10 0

SETS OF APPARATUS FOR PRODUCING THE ELECTRIC LIGHT.

No. 1.

Small-size Electric Lamp, with Reflector (Fig. 17) £2 5 0

20 quart Bunsen Cells, at 6/6 each 6 10 0

2 Varnished Oak Trays, at 10/6 each 1 1 0

Carbon Rod per foot 1/6, per dozen feet 0 10 0

No. 2.

Medium-size Automatic Electric Lamp, without Reflector (Fig. 19) ... 7 10 0

Parabolic Reflector, extra 2 2 0

30 quart Bunsen's Cells, at 6/6 each 9 15 0

3 Varnished Oak Trays, at 10/6 each 1 11 6

Carbon Rod per foot 1/6, per dozen feet 0 15 0

No. 3.

Large Automatic Electric Lamp (Fig. 19) 9 9 0

Parabolic Reflector, extra 2 2 0

50 quart Bunsen's Cells, at 6/6 each 16 5 0

5 Varnished Oak Trays, at 10/6 each 2 12 6

Carbon Rod per foot 1/6, per dozen feet 0 15 0

SPECTRUM APPARATUS FOR THE MICROSCOPE.

Price of the Micro-Spectroscope, complete with Micrometer £8 5 0

The Sorby-Browning Micro-Spectroscope, with rack-motion to eye-piece (Fig. 24), without Micrometer 6 0 0

Browning's Bright-Line Micrometer, for measuring the position of bright lines in Spectra, adapted to the above (Fig. 24) 2 5 0

Case for Spectroscope, with Racks for Cells and Tubes	£0 15 0
Sorby's Tubes	per dozen 0 2 6
Sorby's Wedge Cells	per dozen 0 6 0
Specimens in Sealed Tubes for showing the Bands, each	0 1 6

For List of Specimens for the Micro-Spectroscope, see page 49.

Price of the Specimens—Class I., Class II., 10/6; Class III., 6/6.

The Amateur's Micro-Spectroscope, with Achromatic Lens and Reflecting Prism, to show Two Spectra at the same time, for the purpose of comparison...	2 15 0
Mahogany Case for the above	0 5 0

TO LECTURERS ON SCIENCE.

JOHN BROWNING begs to announce that he has prepared, with peculiar care, a great number of Diagrams, principally Photographs, to illustrate recent discoveries in Spectrum Analysis and other branches of Observational Astronomy. These Slides can be had either plain or exquisitely coloured.

Prices, Plain	£0 3 6
Coloured	from 4/6 to	0 10 6
Photographs of Microscope Objects for the Lantern, each	0 3 6

For List of Subjects, see end of Spectroscope Catalogue.

DENSE GLASS PRISMS.

Prisms of extra dense flint, of very superior quality, of 45 or 60 degrees, with accurate plane surfaces, $\frac{3}{4}$ inch	£0 15 0
Prisms, 1 in., 20/; $1\frac{1}{2}$ in., 30/; $1\frac{1}{2}$ in., 60/; $2\frac{1}{2}$ in., 90/; $2\frac{3}{4}$ in.	6 0 0
Prisms, 3 in., £15; 4 in. by 3 in.	30 0 0
Prisms of the above sizes, of the densest flint made, 25 per cent. extra.					

BISULPHIDE OF CARBON PRISMS.

Bisulphide of Carbon Prisms, large size	£0 15 0
Bisulphide of Carbon Prisms, extra large size	0 18 6
Bisulphide of Carbon Prisms, with Parallel Glass Sides, Browning's improved method of Mounting in Metal Frames	each	1 15 0

Prisms of other Angles to order.

HOLLOW GLASS PRISMS.

Hollow Glass Prisms, with movable sides in metal frames, for taking the refractive index or dispersive power of a liquid, angles of 60°, $1\frac{1}{4}$ in. faces, to replace the usual dense glass prism on a Spectroscope at pleasure	each £0 18 6
Hollow Glass Prisms, with perfectly plane and parallel sides, without metal frames, warranted to give the finest definition	from £3 3/ to	6 6 0

SPECTROGRAPHS.

Beckley's Spectrograph, for Mapping out Spectra (Fig. 26)	£10	0	0
Beckley's Spectrograph, larger...	15	0	0
Colonel Campbell's Automatic Micrometric Spectrograph (Fig. 27)	16	10	0

SUNDRY SPECTROSCOPIC APPARATUS.

Hollow Cells, with one side formed of a Prism, for holding Solutions for examining Absorption Bands	£1	1	0
Large ditto, for projecting Spectra on Screen	1	11	6
Extra power Eye-pieces	12/6 to	1	0	0
Bunsen's Burners	3/6 to	0	5	0
Adjusting Clip, on stand, to hold platinum wire	0	3	6
Browning's Improved Spectroscope Lamp, containing burner and clip on a single stand, complete (Fig. 28)	0	12	6
Brass Stand, superior finished ditto	1	15	0
Leyden Jars	from 3/6 to	2	2	0
Insulated Spark Apparatus, on brass stand, with 2 Dischargers, for obtaining the Spectra of Metals and Gases	1	18	6
Set of 13 chemically pure Metals, in Mahogany Cabinet, for Spectrum experiments...	0	18	6
Vacuum Tubes, prepared for showing the beautiful Spectra of various Gases—Nitrogen, Hydrogen, Oxygen, Carbonic Acid, Ammonia, Sulphuric Acid, Olefiant, Chlorine, Bromine, Iodine, Coal Gas, Æther Vapour, Turpentine Vapour, Petroleum Oil Vapour, and Water Vapour	each, 5/6, 7/6, and	0	8	6
Set of Salts best adapted for showing Chemical Spectra, stoppered bottles, in case	0	7	6

Metallic Thallium and other Chemicals to order.

Platinum Wire, for use with the Spectroscope	per foot	0	1	0
Plucker's Tube Holder, for holding a single Plucker's Tube		1	5	0
Plucker's Tube Holder, for holding 7 tubes	3	3	0
Insulated Spark Apparatus, on brass stand, for obtaining the Spectra of Metals	1	5	0
Lockyer's Insulated Spark Apparatus, with vertical and horizontal rack motions	1	15	0
Lockyer's Insulated Spark Apparatus, with 7 Dischargers, for obtaining the Spectra of 7 Metals without altering the Apparatus (as Fig. 29)...				4	0	0
Lockyer's Insulated Spark Apparatus, with 14 Dischargers (Fig. 29)	...			6	10	0
Reflecting Mirror for following the motion of the Sun, so as to investigate or map the Solar Spectrum. This is a rectangular mirror of large size, mounted on an axis, and provided with endless screw motions in altitude and azimuth, and Hook's joints, with long handles, which can be carried to the foot of the Spectroscope when the mirror is placed outside of a window. Price	8	10	0

The following Prices are Nett for Cash; Half-price allowed for returned Packages, if Carriage Paid.

Orders should be accompanied by a Remittance.

REDUCED LIST OF PRICES OF MIRRORS, Mounted and Unmounted.

JANUARY, 1883.

SILVERED GLASS TELESCOPES AND SPECULA.

SILVERED GLASS SPECULA, UNMOUNTED.

WITHOUT CELLS.

The performance of these Specula will be guaranteed; they will bear a power of 100 to the inch on suitable objects, and under favourable conditions of the atmosphere.

Speculum $3\frac{1}{2}$ inch diameter, about 3 ft. focus	£2 15 0
„ $4\frac{1}{2}$ „ „ 5 „	4 0 0
„ $6\frac{1}{2}$ „ „ 5 ft. or 6 ft. focus	6 15 0
„ $8\frac{1}{2}$ „ „ 5 „ 8 „	14 10 0
„ $9\frac{1}{4}$ „ „ 6 „ 8 „	20 0 0
„ $10\frac{1}{4}$ „ „ 6 „ 9 „	30 0 0
„ $12\frac{1}{4}$ „ „ 6 „ 10 „	40 0 0
„ 13 „ „ 9 „ 10 „	50 0 0
„ 15 „ „ 10 „	65 0 0
„ 18 „ „ 15 „	100 0 0

PRICES OF

SILVERED GLASS SPECULA ASTRONOMICAL TELESCOPES, ON ALT-AZIMUTH STANDS.

$3\frac{1}{2}$ inch Speculum, 3 ft. focus, mounted in metal, on metal alt-azimuth stand, with two eye-pieces, 50 to 150	£10 0 0
$4\frac{1}{2}$ inch Speculum, 5 ft. focus, mounted on a stand, which can be changed from alt-azimuth to parallactic, so that the stars can be followed with one motion, with endless driving screw, and Hook's joint and two eye-pieces, 100 to 200	22 0 0
$6\frac{1}{2}$ inch Speculum, 6 ft. focus, on alt-azimuth stand, with quick and slow fine screw motions, and three eye-pieces, 100 to 450	33 0 0
$8\frac{1}{2}$ inch Speculum, 8 ft. focus, mounted as above, with three eye-pieces, 100 to 500	44 0 0
$9\frac{1}{4}$ inch Speculum, 8 ft. focus, as above, with four eye-pieces, 100 to 600	55 0 0
$10\frac{1}{4}$ inch Speculum, 9 ft. focus, ditto	66 0 0

SILVERED GLASS SPECULA ASTRONOMICAL TELESCOPES.

EQUATORIALLY MOUNTED IN A SUPERIOR MANNER.

$4\frac{1}{2}$ inch Speculum, 5 ft. focus, equatorially mounted (angle for latitude to order), with 6 inch hour circle reading to 5 seconds, and declination circle reading to 1 minute, two eye-pieces, 100 and 300 (Fig. 7)	£45 0 0
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6½ inch Speculum, 6 ft. focus, with 12 inch hour circle reading to 5 seconds, and declination circle to 1 minute, three eye-pieces, 100 to 450, rotating hour circle	£85	0	0
8½ inch Speculum, 8 ft. focus, mounted as above	112	0	0
9½ inch Speculum, 8 ft. focus, with four eye-pieces, 100 to 600	145	0	0
10½ inch Speculum, 9 ft. focus	170	0	0
12½ inch Speculum, with extra eye-pieces	225	0	0
13 inch Speculum, with ten eye-pieces, including Achromatics and Kellner	335	0	0
15 inch Speculum, 10 ft. focus, with 16 inch hour circle, reading to five seconds, and declination circle to one minute, with three Huyghenian eye-pieces, one Kellner, and six Browning's improved Achromatic eye-pieces, powers ranging from 60 to 600 diameters. Position Micrometer, with two verniers divided on silver, and reading to single minutes, with clock-work driving apparatus complete... ..	600	0	0
Clockwork Driving Apparatus, to 8½, 9½, or 10½ inches	38	0	0
Ditto ditto 12½ or 13 inches... ..	48	0	0

SILVERING GLASS SPECULA.

4½ inches	£0	5	0	13 inches	£1	10	0
6½ „	0	10	0	15 „	2	0	0
8½ „	0	12	6	Silvering Plane Mirrors ...	0	2	6
9½ „	0	15	0	Silvering above 2½ inches			
10½ „	1	0	0	in the minor axis ...	0	3	6
12½ „	1	5	0				

All Charges incurred for Carriage will be extra.

ASTRONOMICAL EYE-PIECE—HUYGHENIAN CONSTRUCTION.

Nos. 1 and 2, magnifying 65 and 85	£0	15	0
„ 3, 4, and 5, „ 125, 200, and 250	1	0	0
„ 6 „ 400	1	5	0
„ 7 „ 600	1	10	0

ACHROMATIC EYE-PIECES.

These Eye-pieces have a rather limited field, but their performance with reflecting telescopes particularly on planets, is very superior to Huyghenian.

A Magnifying 86	£1	2	6	E Magnifying 306	£1	15	0
B „ 144	1	10	0	F „ 450	2	0	0
C „ 208	1	15	0	G „ 600	2	5	0
D „ 250	1	15	0	H „ 840	2	10	0

LARGE FIELD EYE-PIECES.

Very low power Comet Eye-piece, magnifying 35	£1	0	0
Kellner Eye-piece, with field of 50 minutes, for clusters or nebulae, magnifying 60... ..	1	7	6
Ditto, with field of 35 minutes, for clusters, nebulae, or the moon, magnifying 85	1	7	6
Day-power Eye-pieces, erect	£1	0	0 to 1 10 0

The power of all the Eye-pieces has been calculated on an object-glass, or mirror of six feet focus.

SOLAR EYE-PIECES.

Single reflecting Prism, mounted, for viewing the sun or moon, with two shade heads, in mahogany case	£3	10	0
Ditto, with two Prisms, arranged for single reflections, for viewing the sun only	8	0	0

TRANSIT EYE-PIECES.

Transit Eye-picces, with fine webs or wires	£1	5	0
Ditto, with 7 webs and higher powers	£1	10	0	to	1 15 0

PARALLEL WIRE MICROMETER.

Micrometer for measuring to a fraction of a second £6 to 11 11 0

POSITION MICROMETER.

[illegible]

Extra Eye-pieces, 15/ each.

Browning's Double-image Micrometer	\$ 10 00
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BARLOW'S LENS.

This is an achromatic combination of a negative focus ; on inserting it behind any eye-piece (that is, between the eye-piece and the object-glass) the power of the eye-piece is increased from one-third to one-half ; at the same time the introduction of this Lens, especially when using Huyghenian eye-pieces, greatly improves the performance of reflecting telescopes, especially on bright stars.

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fraction for intensifying the marks on the Moon or Planets... ..

NEUTRAL TINT WEDGES, for observing the Sun	1	2	0
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DIVIDED LENS DYNAMETER, for measuring accurately the power of Eye-pieces, new movement for separating the Lenses ...

power of Eye-pieces, new movement for separating the Lenses ... 4 4 0

ASTROMETER.

Knobel's Astrometer, a simple and efficient instrument for determining
Star Magnitudes £12 10 0

PERFECT PLANES, UNMOUNTED.

PERFORMANCE UNDER ANY POWER GUARANTEED.

1 inch in the minor axis of the Ellipse	\mathcal{L}^1	0	0
$1\frac{1}{2}$ " " "	1	10	0
2 " " "	2	0	0
$2\frac{1}{2}$ " " "	2	10	0
3 " " "	3	5	0

Prepared Pad for polishing Specula, in bottle, 2s. 6d.

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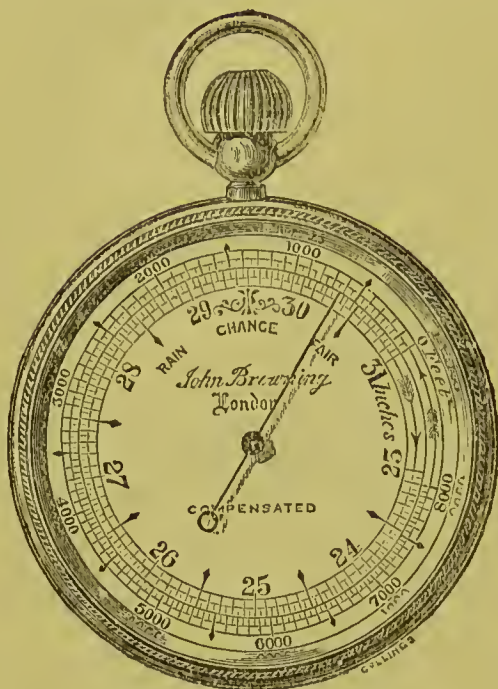
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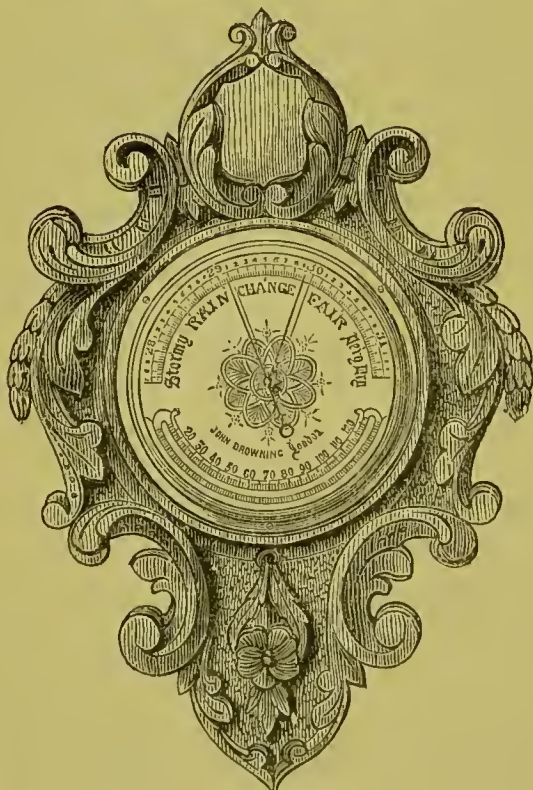
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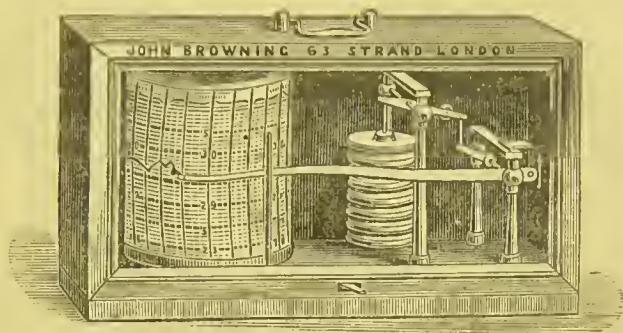
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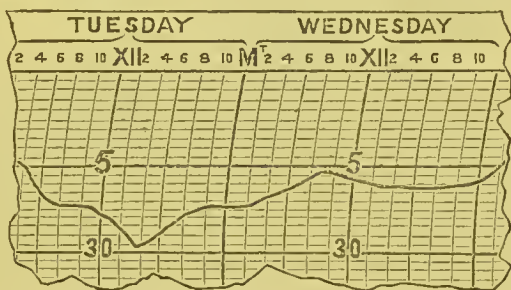
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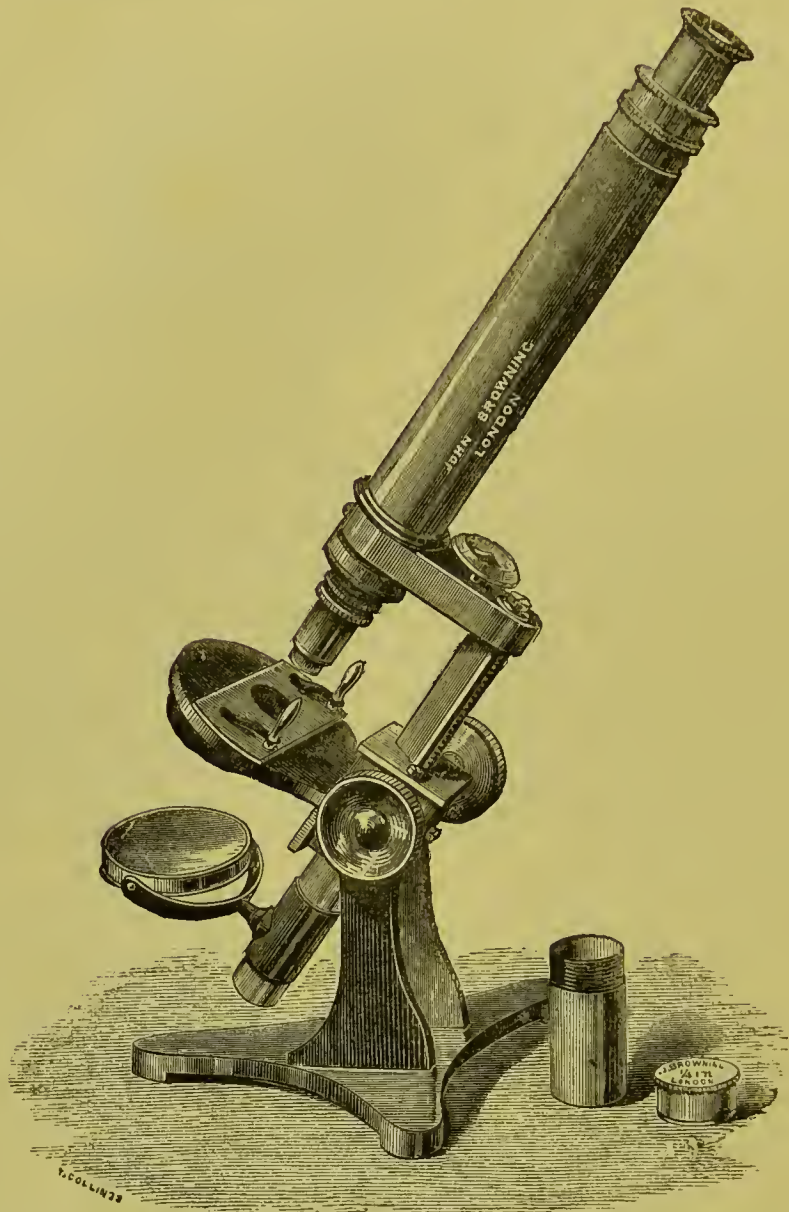
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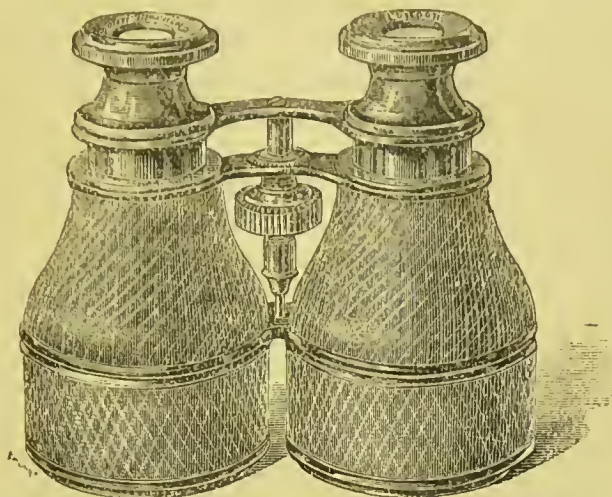
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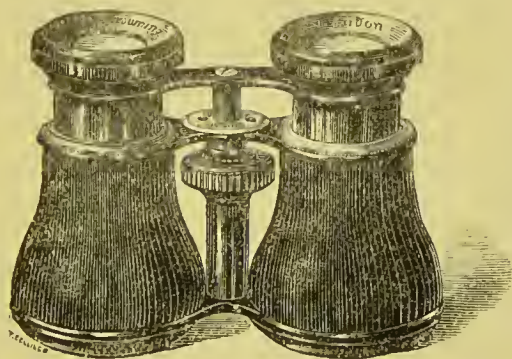
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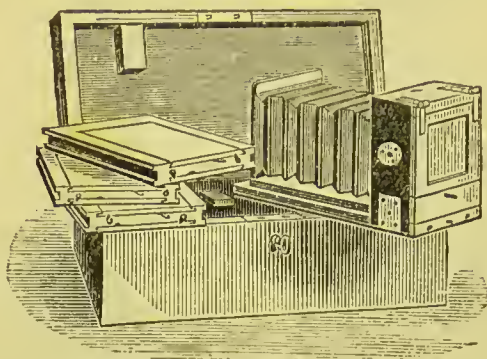
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